Atmos. Chem. Phys. Discuss., 12, 14269–14327, 2012 www.atmos-chem-phys-discuss.net/12/14269/2012/ doi:10.5194/acpd-12-14269-2012 © Author(s) 2012. CC Attribution 3.0 License.



This discussion paper is/has been under review for the journal Atmospheric Chemistry and Physics (ACP). Please refer to the corresponding final paper in ACP if available.

## Model uncertainties affecting satellite-based inverse modeling of nitrogen oxides emissions and implications for surface ozone simulation

J.-T.  $Lin^1$ , Z.  $Liu^2$ , Q. Zhang<sup>3</sup>, H.  $Liu^4$ , J. Mao<sup>5,6</sup>, and G. Zhuang<sup>7</sup>

<sup>1</sup>Laboratory for Climate and Ocean-Atmosphere Studies, Department of Atmospheric and Oceanic Sciences, School of Physics, Peking University, Beijing 100871, China

<sup>2</sup>School of Earth and Atmospheric Sciences, Georgia Institute of Technology, Atlanta, Georgia 30332, USA

<sup>3</sup>Center for Earth System Science, Tsinghua University, Beijing, China

<sup>4</sup>National Institute of Aerospace, Hampton, VA 23666, USA

<sup>5</sup>Program in Atmospheric and Oceanic Sciences, Princeton University, Princeton, NJ 08542, USA

<sup>6</sup>Geophysical Fluid Dynamics Laboratory/National Oceanic and Atmospheric Administration, Princeton, NJ 08542, USA

<sup>7</sup>Center for Atmospheric Chemistry Study, Department of Environmental Science and Engineering, Fudan University, Shanghai, 200433, China



Received: 22 April 2012 - Accepted: 29 May 2012 - Published: 7 June 2012

Correspondence to: J.-T. Lin (linjt@pku.edu.cn)

Published by Copernicus Publications on behalf of the European Geosciences Union.



#### Abstract

Errors in chemical transport models (CTMs) interpreting the relation between spaceretrieved tropospheric column densities of nitrogen dioxide ( $NO_2$ ) and emissions of nitrogen oxides ( $NO_x$ ) have important consequences on the inverse modeling. They

- <sup>5</sup> are however difficult to quantify due to lack of adequate in situ measurements, particularly over China and other developing countries. This study proposes an alternate approach for model evaluation over East China, by analyzing the sensitivity of modeled NO<sub>2</sub> columns to errors in meteorological and chemical parameters/processes important to the nitrogen abundance. As a demonstration, it evaluates the nested version
- <sup>10</sup> of GEOS-Chem driven by the GEOS-5 meteorology and the INTEX-B anthropogenic emissions and used with retrievals from the Ozone Monitoring Instrument (OMI) to constrain emissions of  $NO_x$ . The CTM has been used extensively for such applications. Errors are examined for a comprehensive set of meteorological and chemical parameters using measurements and/or uncertainty analysis based on current knowl-
- edge. Results are exploited then for sensitivity simulations perturbing the respective parameters, as the basis of the following post-model linearized and localized first-order modification. It is found that the model meteorology likely contains errors of various magnitudes in cloud optical depth, air temperature, water vapor, boundary layer height and many other parameters. Model errors also exist in gaseous and heterogeneous re-
- actions, aerosol optical properties and emissions of non-nitrogen species affecting the nitrogen chemistry. Modifications accounting for quantified errors in 10 selected parameters increase the NO<sub>2</sub> columns in most areas with an average positive impact of 22 % in July and 10 % in January. This suggests a possible systematic model bias such that the top-down emissions will be overestimated by the same magnitudes if the model is
- <sup>25</sup> used for emission inversion without corrections. The modifications however cannot account for the large model underestimates in cities and other extremely polluted areas (particularly in the north) as compared to satellite retrievals, likely pointing to underestimates of the a priori emission inventory in these places with important implications



for understanding of atmospheric chemistry and air quality. Post-model modifications also have large impacts on surface ozone concentrations with the peak values in July over North China decreasing by about 15 ppb. Individually, modification for the uptake of the hydroperoxyl radical on aerosols has the largest impact for both NO<sub>2</sub> and ozone, followed by various other parameters important for some species in some seasons.

<sup>5</sup> followed by various other parameters important for some species in some seasons. Note that these modifications are simplified and should be used with caution for error apportionment.

### 1 Introduction

Anthropogenic emissions of nitrogen oxides  $(NO_x \equiv NO + NO_2)$  in China have been <sup>10</sup> growing at an average rate of about 8 % per year (Richter et al., 2005; Zhang et al., 2007; Lamsal et al., 2011; Lin and McElroy, 2011) over the past decade, disrupted temporarily by various socioeconomic events such as the economic downturn, Olympics and Chinese New Year (Mijling et al., 2009; Lin et al., 2010a; Lin and McElroy, 2011). The rising emissions have attracted tremendous attention concerning the con-<sup>15</sup> sequences on tropospheric chemistry, air pollution, climate forcing and acid deposition over both source and downwind regions (e.g., Streets et al., 2003; Richter et al., 2005; Martin et al., 2006; Zhang et al., 2007; Stavrakou et al., 2008; Lin et al., 2010a; Lin, 2012).

Simulations of chemical transport models (CTMs) are employed extensively with
space based measurements of vertical column densities of tropospheric nitrogen dioxide (NO<sub>2</sub>) to constrain emissions and variability of NO<sub>x</sub> in China from the top-down perspective (e.g., Jaeglé et al., 2005; van der A et al., 2006; Lin and McElroy, 2011; Lin, 2012). Such top-down constraint is a useful supplement to bottom-up estimates which are subject to uncertainties in emission factors and emission activities (Streets
et al., 2003; Zhang et al., 2007, 2009b). It is however affected by errors not just in satellite retrievals but also in CTMs simulating the fate of nitrogen emitted into the troposphere (Martin et al., 2006; Zhao and Wang, 2009; Lin et al., 2010b; Lin, 2012).



Errors in model simulations are estimated often from ground-based and aircraft measurements of nitrogen, mostly in a short time period, over the developed countries or over remote/downwind regions (e.g., Hudman et al., 2007; Brinksma et al., 2008; Boersma et al., 2009; Lamsal et al., 2010; Lin and McElroy, 2010). Conclusions drawn

- <sup>5</sup> are adopted for China and other countries/regions where in situ measurements are not available (Jaeglé et al., 2005; Martin et al., 2006; Boersma et al., 2008; Zhao and Wang, 2009; Lin et al., 2010b; Wang et al., 2012). However, the magnitude and causes of model errors may be region and time dependent, resulting in uncertainties in the applicability of findings from model evaluation limited in time and space. For example,
- errors in model meteorology may depend on locations and seasons; and errors in heterogeneous reactions on aerosol surfaces are important mainly at locations with high aerosol loadings (see Sects. 4 and 5). Even for areas with nitrogen observations, it is still difficult to evaluate the model capability in relating emissions and atmospheric abundances based on measured concentrations alone without accurate information
- on emissions. Lin and McElroy (2010) evaluated the modeled vertical mixing in the planetary boundary layer (PBL) over the Northeast USA using aircraft observations. They noted that modeled concentrations of NO<sub>2</sub> are much lower than observed values throughout the PBL, which may or may not be caused entirely by emission errors.

The present study proposes an alternate approach for model evaluation over China,

- <sup>20</sup> by analyzing the sensitivity of simulated NO<sub>2</sub> columns to model meteorological and chemical parameters of importance to the nitrogen chemistry (Fig. 1). A large number of sensitivity simulations are conducted to evaluate the impacts of errors in model meteorological fields constrained by ground and space based measurements. The model chemistry is evaluated with additional simulations addressing the embedded
- errors/uncertainties based on current understanding. Improvements on the estimate of NO<sub>2</sub> columns are achieved then by linearized and localized first-order modifications accounting for quantified errors in model meteorological and chemical parameters. The method is useful for China due particularly to lack of adequate and timely measurements of nitrogen constituents for more direct model evaluation.



As a demonstration, this study evaluates tropospheric NO<sub>2</sub> columns simulated by the nested GEOS-Chem model (Chen et al., 2009) in use of inferring emissions of NO<sub>x</sub> from NO<sub>2</sub> columns retrieved from the Ozone Monitoring Instrument (OMI). GEOS-Chem has been used extensively for such top-down constraints over a variety of tem-<sup>5</sup> poral and spatial scales (Jaeglé et al., 2005; Martin et al., 2006; Boersma et al., 2008; Zhang et al., 2008; Lin et al., 2010b; Lamsal et al., 2011; Lin and McElroy, 2011; Lin,

- 2012; Wang et al., 2012). Findings from the analysis of GEOS-chem simulations may be applicable to some extent to other models with different meteorological data and/or chemical schemes. The analysis is focused in East China (101.25°–126.25° E, 19°–
- <sup>10</sup> 46° N) concerning the extensive inverse modeling studies addressing its significant and rapidly increasing emissions of NO<sub>x</sub> in recent years (e.g., Richter et al., 2005; Zhang et al., 2007; Stavrakou et al., 2008; Lin, 2012). Simulations for January and July 2006 are both evaluated to account for potential seasonal dependence of model errors.
- Errors in model meteorological and chemical parameters also have significant im-<sup>15</sup> plications for other model applications. The impacts on surface ozone simulation are analyzed in the present study to further demonstrate the importance of explicitly accounting for model errors in respective applications.

The remaining sections are organized as follows. Section 2 presents meteorological observations from the ground networks and satellite remote sensing, as well as

- space based measurements of aerosol optical depth (AOD) and NO<sub>2</sub> columns. Section 3 presents the CTM simulations and a brief comparison with satellite retrievals. Sections 4 and 5 evaluate the impacts on simulated NO<sub>2</sub> columns of potential errors in model meteorological and chemical parameters, respectively, by employing a large set of sensitivity simulations. The test simulations are conducted based current under-
- standing of the nitrogen chemistry, as shown in Fig. 1. Section 6 makes an attempt to modify the simulated NO<sub>2</sub> columns on the first order accounting for quantified errors in model meteorology and chemistry. Implications of potential meteorological and chemical errors for simulations of surface ozone are discussed in Sect. 7. Section 8 summarizes the present study.



#### 2 Ground and space measurements

#### 2.1 Meteorological observations

Ground measurements for January and July 2006 are taken from the National Oceanic and Atmospheric Administration (NOAA) National Climatic Data Center (NCDC) Inte-

- <sup>5</sup> grated Surface Hourly (ISH) dataset (DS3505) for a variety of meteorological parameters, including air temperature, relative humidity (RH), surface air pressure, wind speed at 10 m, and precipitation. The ISH dataset combines measurements from various networks publically available for weather and climate analyses. It was used by Lin and McElroy (2011) to analyze the day-to-day variation of meteorology in January 2009 of relevance to changes in NO<sub>2</sub> columns retrieved from space and simulated by GEOS-Chem. Here meteorological data are taken from a total of 284 stations across East China. They are available every three hours (starting from 00:00 UTC) with the exception of daily precipitation data.
- Space measurements of tropospheric water vapor path, cloud fraction and cloud optical depth (COD) are taken from the International Satellite Cloud Climatology Project (ISCCP) D2 dataset. The ISCCP dataset is used widely in studies of clouds and the hydrological cycle. It provides monthly mean data on the 2.5° long × 2.5° lat grid for time of day from 00:00 to 21:00 UTC at 3-h intervals for cloud fraction and COD, and monthly mean data for water vapor path. Here data for January and July 2006 are sampled at the locations of ground meteorological stations for comparison with the GEOS-5 meteorology.

#### 2.2 NO<sub>2</sub> columns from OMI

25

Retrievals of NO<sub>2</sub> columns are taken from the DOMINO product version 2 (Boersma et al., 2011). The processing of DOMINO-2 is presented in Lin (2012), including but not limited to the conversion from level 2 to level 3 monthly data and an analysis of retrieval



14276

errors. The level 3 data are derived on the  $0.25^{\circ}$  long ×  $0.25^{\circ}$  lat grid for January and July 2006.

#### 2.3 MODIS AOD

To evaluate modeled optical properties of aerosols, measurements of aerosol optical depth (AOD) at 550 nm are taken from the MODIS/Aqua level 2 collection 5.1 product. Data for January and July 2006 are sampled at the locations of ground meteorological stations for comparison with GEOS-Chem simulations. For a given station, valid MODIS data in each day available within 0.25° of the station are averaged to obtain the daily value.

#### **3 GEOS-Chem simulation and comparison with OMI retrievals**

This study evaluates the nested model for Asia (Chen et al., 2009) of GEOS-Chem (version 8-03-02; http://wiki.seas.harvard.edu/geos-chem/index.php/MainPage). The model is driven by the assimilated meteorological fields of GEOS-5 taken from the National Aeronautics and Space Administration (NASA) Global Modeling and Assimilation Office (GMAO). It is run with the full O<sub>x</sub>-NO<sub>x</sub>-CO-VOC-HO<sub>x</sub> chemistry and online 15 calculation for various aerosols (sulfate, nitrate, ammonium, black carbon, primary organic carbon, sea salt and dust). The nested model has a horizontal resolution of  $0.667^{\circ}$  long ×  $0.5^{\circ}$  lat with 47 vertical layers; and each of the lowest 10 layers is about 130 m thick. Its lateral boundary conditions are updated every 3 h from a respective global simulation at 5° long × 4° lat. Other model setup is described in more detail 20 in Lin (2012), including emissions and their temporal variability. In particular, anthropogenic emissions in Asia are taken from the INTEX-B inventory as representative of 2006 (Zhang et al., 2009b) for various species including NO<sub>x</sub>, carbon monoxide (CO), non-methane volatile organic compounds (VOC), sulfur dioxide (SO<sub>2</sub>), black carbon (BC) and primary organic carbon (OC). 25



For all simulations (see Table 1), modeled vertical distribution of NO<sub>2</sub> in each day is sampled at the time with valid retrievals, regridded to 0.25° long × 0.25° lat, and applied with the averaging kernel (AK) from DOMINO-2. The daily data are averaged then to obtain monthly mean values in January and July 2006 for the present analysis.

- NO<sub>2</sub> columns derived from the base simulation (Case 1 in Table 1) are compared in 5 detail to retrieved values by Lin (2012). Averaged over East China, model NO<sub>2</sub> columns are lower than the retrievals by about 20% in July and by 36% in January. The underestimate is much more significant in extremely polluted areas such as cities and locations with large point sources, particularly in the north in January: model values are only
- about 20-30 % of retrievals. Results are similar from the comparison at the 284 meteo-10 rological stations showing a slope of about 0.5 in both months under the reduced major axis (RMA) regression (Fig. 2a, b). Lin (2012) found that total anthropogenic emissions of NO<sub>v</sub> in East China inferred from DOMINO-2 are larger than the INTEX-B inventory by about 40%. These differences are attributable in part to errors in model simulations,
- as analyzed in the following sections. 15

20

#### 4 Sensitivity of GEOS-Chem simulations to meteorological parameters

Errors in the GEOS-5 meteorological fields have consequences on a variety of atmospheric processes affecting the abundance of tropospheric NO<sub>2</sub> simulated by GEOS-Chem. The GEOS-5 meteorology is evaluated in the Supplement with Figs. S1-S8 using observational data from ISH and ISCCP. Parameters analyzed include air temperature, RH, tropospheric water vapor path, air pressure, wind speed, cloud fraction, COD and precipitation. The analysis is emphasized in the daytime, particularly at midday when the lifetime of NO<sub>v</sub> is shortest and has the largest impact on its abundance at the overpass time of OMI (i.e., in the early afternoon).

In this section, sensitivity simulations are conducted to evaluate the impacts of errors 25 in air temperature, water vapor content and COD on modeled NO<sub>2</sub> columns, by adjusting the respective parameters in GEOS-Chem (see Fig. 1 and Table 1). Also analyzed



are the impacts of potential errors in convection (and resulting lightning emissions of  $NO_x$ ) and vertical mixing in the PBL.

Errors in RH affect wet deposition of various species and hygroscopic growth of aerosols. These factors are accounted for implicitly with adjusted air temperature and water vapor as the CTM re-calculates RH on the fly. Impacts of errors in 10 m wind speed on modeled NO<sub>2</sub> columns are mainly on the local scale due to the relatively short lifetime of NO<sub>x</sub>. Also, biases in wind speed are likely limited to altitudes close to the ground where small-scale variability is not captured by GEOS-5, as expected for an assimilation product (Pryor and Barthelmie, 2011). At higher altitudes, wind speed is better constrained, since the large-scale circulation is reasonably reproduced by GEOS-5 (as suggested by the evaluation for surface air pressure in the Supplement). Errors in precipitation are small averaged over East China (see Supplement). They are also not very important for the simulation of NO<sub>x</sub> whose wet scavenging is relatively weak. Therefore the impacts of these parameters on modeled NO<sub>2</sub> columns are not

evaluated explicitly in the present study.

#### 4.1 Air temperature

20

Errors in air temperature affect the calculation of gaseous and heterogeneous reactions, deposition and biogenic emissions of volatile organic compounds (VOC), with consequences on modeled  $NO_2$  columns. Their overall impact is evaluated here by sensitivity simulations perturbing air temperature in GEOS-5 (Cases 2–5 in Table 1), based on analysis of temperature in the Supplement.

The first simulation increases the nighttime temperature by 1 °C and decreases the daytime temperature by 2 °C in the lowest 10 model layers (about 0–1.3 km above the ground) (Case 2 in Table 1). The adjustment extends to the lower troposphere to ac-<sup>25</sup> count for potential errors that may be transported from near the ground to the entire PBL. It also affects the ground temperature with consequences on modeled biogenic emissions. As a result, modeled NO<sub>2</sub> columns in July are reduced by 0–4 % over most regions with enhancements by up to 10 % in parts of the southwest (Fig. S9a). This



is mainly a result of enhanced formation of nitric acid (through enhanced  $NO_2 + OH$  reaction, a termolecular reaction with negative dependence on temperature) compensated by reduced production of organic nitrates (through reduced emissions of VOC). For January, model  $NO_2$  columns are reduced by up to 8 % in some areas (Fig. S9b).

- <sup>5</sup> The daytime biases in air temperature exceed 5 °C in parts of East China (Fig. S1). Therefore another sensitivity simulation as an extreme case increases the daytime temperature by 5 °C with a reduction by 1 °C at night (Case 3 in Table 1). It reduces model NO<sub>2</sub> columns by 4–10 % in many areas of East China for both months (Fig. 3). In part of the southwest, model NO<sub>2</sub> columns are enhanced by up to 16 % in July. Assuming
- <sup>10</sup> a weaker effect of PBL mixing on temperature errors, the temperature adjustment is constrained to the lowest four model layers (i.e., about 0–500 m above the ground) in a further simulation (Case 4 in Table 1). This tends to slightly increase the NO<sub>2</sub> columns relative to when the temperature is adjusted for the lowest 10 layers (Fig. S9e, f versus Fig. 3).
- <sup>15</sup> The nighttime temperature errors are not important for simulation of NO<sub>2</sub> columns in the afternoon. This is suggested by an additional simulation increasing the nighttime temperature by 5 °C with no changes in the daytime (Case 5 in Table 1). Specifically, the sensitivity test results in insignificant changes in model NO<sub>2</sub> columns in both months: within 2 % for July and within 4 % in most areas for January (not shown).

#### 20 4.2 Water vapor content

Water vapor content in the troposphere affects the yield of hydroxyl radical (OH) from ozone photolysis and consequently the loss rate of NO<sub>x</sub> via OH oxidation. It also affects the hygroscopic growth of hydrophilic aerosols and resulting scattering efficiency for solar radiation. Based on analysis of the GEOS-5 water vapor path in the Supplement, <sup>25</sup> two test simulations are conducted to quantify the sensitivity of model NO<sub>2</sub> columns to errors in water vapor content in GEOS-5. In particular, the water vapor content is scaled up and down, respectively, by a factor of 130 % in all tropospheric layers (Cases 6–7 in Table 1). Increasing the water vapor content results in 4–10 % reductions in



 $NO_2$  columns across East China in July with a smaller impact for January (Fig. 4a, b). Decreasing the water vapor content has opposite impacts on model  $NO_2$  columns for both months (Fig. 4c, d).

### 4.3 Cloud optical depth

- <sup>5</sup> Model errors in COD affect the calculation of photolysis for a variety of gaseous tracers (Liu et al., 2009). The COD in GEOS-5 is lower than 50% of the ISCCP values in most areas of East China for both January and July (Fig. S7). Here four sensitivity simulations are performed to evaluate the effect of COD on model photolysis and resulting NO<sub>2</sub> columns, by multiplying the COD by a factor of 0%, 50%, 150% and
- <sup>10</sup> 200 %, respectively (Cases 8–11 in Table 1). The resulting impacts on NO<sub>2</sub> columns are significant especially in the eastern and southern provinces in July (Figs. 5, S10a– h). In particular, doubling the COD in July enhances modeled NO<sub>2</sub> columns by 20 % or more in many places (Fig. 5), by decreasing solar radiation in the lower troposphere allowed for photodissociation of ozone and NO<sub>2</sub> and formation of OH.
- At a given COD, the altitude of clouds affects the vertical distribution of radiation and photolytic activities. In lack of measurements of cloud vertical distribution, the effect of potential model errors is evaluated by a sensitivity simulation distributing the COD evenly in all vertical layers within the troposphere (Case 12 in Table 1). For July, most clouds in GEOS-5 are located in the upper troposphere (not shown), thus the
- arbitrary change in cloud vertical profile allows more solar radiation to reach the lower troposphere. This results in a significant increase in the tropospheric OH content with a reduction in NO<sub>2</sub> by up to 16 % (Fig. S10i). For January, most clouds are located in the lower troposphere, especially in the south (not shown); thus the arbitrarily assumed distribution of COD results in slight increases of NO<sub>2</sub> over these regions (Fig. S10j).



#### 4.4 Planetary boundary layer height (PBLH)

Vertical mixing in the PBL of nitrogen and other species affect the non-linear chemical processes in the lower troposphere. Accurate simulation of vertical mixing is important particularly for comparison with satellite retrievals when the altitude-dependent AK (i.e.,

- satellite sensitivity to changes in NO<sub>2</sub> concentrations) is applied to modeled vertical distribution of NO<sub>2</sub>. In lack of direct observations to evaluate the simulation of PBL mixing for East China, the impact of potential errors is analyzed here by adjusting the PBLH, a key parameter constraining the vertical mixing, in GEOS-Chem (Cases 13–18 in Table 1). In addition to the impacts on the nonlinear chemistry, increasing the PBLH results in enhanced vertical mixing which lifts NO<sub>2</sub> to higher altitudes with larger
- AK and thus enhances the  $NO_2$  column (proxy); and decreasing the PBLH has an opposite effect.

The current GEOS-Chem employs a non-local scheme (Holtslag and Boville, 1993) to determine the extent of vertical mixing in the PBL, with the PBLH taken directly from GEOS-5 (Lin and McElroy, 2010). The GEOS-5 PBLH is diagnosed as the height

- of the lowest layer in which the eddy diffusivity is smaller than  $2 \text{ m}^2 \text{ s}^{-1}$  (Global Modeling and Assimilation Office, 2006). Across East China, its values in the afternoon (13:00–15:00 Beijing local time; around the overpass time of OMI) exhibit significant seasonal variation due to changes in static and mechanic instability. In July, the after-
- noon height exceeds 2000 m over the arid/semiarid northwest but is only about 1000 m in the coastal provinces in the east and south (Fig. S11). In January, it reaches 2000 m in parts of Sichuan but is as low as about 750 m in the north (Fig. S11). At night, the PBLH in GEOS-5 is less than 100 m in most days of the two months over China and other continental areas (not shown). Overall, the GEOS-5 PBLH may be overestimated
- in the daytime due to positive biases in surface air temperature enhancing the static instability; while it may be underestimated at night as a result of negative temperature biases. The observation-based analysis for the USA by Liu and Liang (2010) suggested



the nighttime PBLH to be about 200–300 m in all seasons, about 3 times as large as the GEOS-5 values.

To evaluate the importance of PBLH for modeled  $NO_2$  columns, a sensitivity simulation calculates the height on the fly using the non-local scheme (Case 13 in Table 1).

- For July, this increases the afternoon PBLH over the southeast and northwest with reductions over the southwest (Fig. S11). Consequently, model NO<sub>2</sub> columns are enhanced by 0–4% near the northern boundary of China, but are reduced by 0–4% in most areas (Fig. 6a). For January, the derived PBLH in the afternoon is higher than the GOES-5 values by 100–200 m over North China but is lower over the vast west
   (Fig. S11). As a result, model NO<sub>2</sub> columns decrease by 2–12% over most of East
  - China with enhancements by up to 8% in several provinces (Fig. 6b).

A second sensitivity simulation is conducted by decreasing the PBLH by 10 % in the daytime and tripling the height at night to account for their potential biases (Case 14 in Table 1). For July, this enhances model NO<sub>2</sub> columns by 0–6 % over most regions

(Fig. 6c). For January, changes in model NO<sub>2</sub> columns are more inhomogeneous spatially (Fig. 6d). The effects are driven mainly by the enhanced nighttime PBLH, since the adjustment is relatively small for the daytime PBLH. This is confirmed by another simulation tripling the nighttime PBLH with no adjustment in the daytime (Case 15 in Table 1; figures are not shown).

To further test the model sensitivity to PBLH, an additional simulation employs the height data from the GEOS-4 meteorological fields to adjust values in GEOS-5 (Case 16 in Table 1). For a given location and time of day, the ratio of monthly mean GEOS-4 over GEOS-5 PBLH available on the 2.5° long × 2° lat grid is used to scale the high-resolution GEOS-5 data. The GEOS-4 PBLH is in general higher than GEOS-

<sup>25</sup> 5 in the nighttime with closer agreement to the observation-based estimate by Liu and Liang (2010). For the mid-day PBLH, the GEOS-4 data are lower than GEOS-5 over most of East China in January (Fig. S12), resulting in reductions of model NO<sub>2</sub> columns by 0–20% in many areas (Fig. 6f). They however exceed the GEOS-5 values in the



northeast with a consequent increase in model  $NO_2$  columns. In July, the impacts of PBLH adjustments are within 4 % in most areas for model  $NO_2$  columns (Fig. 6e).

Since  $NO_2$  columns from the base simulation are lower than satellite retrievals by about 20% in July and 36% in January averaged over East China, we further test the

- <sup>5</sup> possibility that the differences originate from errors in modeled PBL mixing. As such, the PBLH is scaled up by 130 % and 200 %, respectively, in two sensitivity simulations (Cases 17–18 in Table 1). For July, increasing the PBLH by 30 % enhances the NO<sub>2</sub> columns by 0–10 % over most regions (Fig. S13a). The 100 % increase in PBLH results in enhancements of model NO<sub>2</sub> columns by 10–20 % over a large portion of East Oliver (Fig. 14.) The substance of the balance of the balan
- <sup>10</sup> China (Fig. 14c). The impacts in January are much larger than that for July: model  $NO_2$  columns are enhanced by 10–20% (20–50%) in many areas of the east and south when the PBLH is enhanced by 30% (100%) (Fig. S13b, d).

#### 4.5 Convection and consequent lightning emissions

Convection affects the vertical distributions of NO<sub>x</sub> and other species by transporting pollutants between the lower and upper troposphere. It however is highly parameterized and subject to large uncertainties in current models (Tost et al., 2010). As indicators of convection, the cloud amount is underestimated by GEOS-5 while precipitation is reproduced relatively well on the regional mean basis (see Supplement Sects. S1.6 and S1.7). Therefore it is currently unclear about the magnitude of convection errors

in GEOS-5 and consequent impacts on simulations of GEOS-Chem. Lin (2012) conducted a sensitivity analysis assuming a 50% increase in the convection of NO<sub>2</sub> of anthropogenic origin. They consequently found an increase by 7.5% in anthropogenic NO<sub>2</sub> columns averaged over East China (when the averaging kernel is applied to modeled vertical distribution of NO<sub>2</sub>). Further research is needed to improve the understanding of convection and its representation in the model.

Convection results in lightning flashes and consequent emissions of  $NO_x$ . The importance of lightning emissions is affected also by the yield of nitric oxide (NO) from each flash and its vertical distribution, which processes are not well understood (Sauvage



et al., 2007). The contribution of a given amount of lightning emissions to model  $NO_2$  columns (when the AK is implemented) is found to be about 50% larger than that of anthropogenic emissions averaged over East China (Lin et al., 2010b; Lin, 2012). The inverse modeling by Lin (2012) suggested that GEOS-Chem underestimates lightning emissions in East China by 14% in July 2006 and by 19% for the entire year. The exact causes of the biases are not clear, however.

#### 5 Sensitivity of GEOS-Chem simulations to chemical parameters

5

Originally developed as a global CTM, GEOS-Chem adopts a simplified chemical scheme to reduce the complexness and computational cost for global and regional analyses, as compared to more detailed schemes such as MCM 3.2 (Saunders et al., 2003). It has been suggested that the GEOS-Chem chemical scheme produces relatively consistent results with MCM 3.1 for inorganic gaseous chemistry, taking into account differences in the kinetic data used (JPL versus IUPAC) (Emmerson and Evans, 2009). The differences increase for organic gaseous chemistry (Emmerson and Evans, 2009).

- <sup>15</sup> 2009), as expected for a largely simplified chemical scheme, with potentially important implications for the nitrogen evolution over areas with abundant VOC. In addition, the implementation of heterogeneous reactions and kinetic data differs between current chemical schemes. In particular, GEOS-Chem mainly adopts the kinetic data from the Jet Propulsion Laboratory (JPL) (Sander et al., 2011), differing from the IUPAC data
- (Atkinson et al., 2004, 2006) employed in MCM and some other schemes. Therefore the sensitivity analysis here is concentrated on chemical parameters/processes that have been identified to be important for the nitrogen chemistry in general (rather than specific ally to GEOS-Chem) (see Fig. 1 and Table 1). Results here can thus be applied to some extent to other (and maybe more sophisticated) chemical schemes.



#### 5.1 Rate constant for reaction of OH and NO<sub>2</sub>

5

Reaction of  $OH + NO_2$  producing nitric acid (HONO<sub>2</sub>, or HNO<sub>3</sub>) is the primary sink of  $NO_x$  in the troposphere. Recent lab experiments suggested a second pathway producing pernitrous acid (HOONO) with a branching ratio of 5–15% under atmospheric conditions (Sander et al., 2011). The newly found pathway has not been accounted for explicitly in the current GEOS-Chem model since HOONO is likely unstable and easily converted back to NO<sub>2</sub>. It is excluded from the analysis hereafter.

The current GEOS-Chem adopts the rate constant for the nitric acid branch recommended by the JPL with an uncertainty of 30 % at 298 K (Sander et al., 2011). A recent lab study suggested the rate constant to be about 15 % lower than the JPL value at 298 K without specifying the values at other temperatures (Mollner et al., 2010), in support of some previous studies indicating the current JPL value to be biased high (Okumura and Sander, 2005). Here two sensitivity simulations are performed to evaluate the impact of potential errors in the rate constant on modeled NO<sub>2</sub> columns, by scaling up and down, respectively, the rate constant by a factor of 130 % (Cases 19–20

<sup>15</sup> scaling up and down, respectively, the rate constant by a factor of 130 % (Cases 19–20 in Table 1). Decreasing the rate constant enhances model NO<sub>2</sub> columns by 10–20 % across East China in July and by 0–10 % in January (Fig. 7a, b). Increasing the rate constant has an opposite effect of similar magnitude (not shown).

#### 5.2 Yield of HNO<sub>3</sub> from reaction of NO and hydroperoxyl radical (HO<sub>2</sub>)

An important pathway of NO<sub>2</sub> formation is reaction of NO and HO<sub>2</sub>. Lab experiments by Butkovskaya et al. (2005, 2007, 2009) suggested a small yield of HNO<sub>3</sub> during the reaction resulting in loss of NO<sub>x</sub>. The yield depends on temperature, pressure and water vapor content. It is less than 1 % in dry air, but can be enhanced by up to an order of magnitude in the presence of water vapor (Butkovskaya et al., 2009; Sander et al., 2011). Here, two sensitivity simulations are performed by assuming a constant yield of HNO<sub>3</sub> at 5 % and 1 %, respectively (Cases 21–22 in Table 1). A yield of 5 % reduces model NO<sub>2</sub> columns by 44 % in July and by 12.5 % in January averaged over



East China with much larger impacts in many areas (Fig. 8a, b). With a yield of 1 %, model NO<sub>2</sub> columns are reduced by 15 % in July and by 3.4 % in January averaged over East China (Fig. 8c, d). The yield of  $HNO_3$  and its dependence on meteorological conditions is yet to be confirmed by more lab experiments.

#### 5 5.3 Isoprene chemistry and PAN

Reaction of isoprene with OH is an important process affecting the budget and partitioning of hydrogen oxide radicals ( $HO_x$ ). It was assumed traditionally that the reaction resulted in conversion of OH to peroxy radicals ( $RO_2$ ) and thus depletion of OH in pristine forested areas with large isoprene sources such as the Amazonian region. More recent research, however, have suggested several pathways of OH regeneration through subsequent reactions of isoprene oxidation products (Butler et al., 2008; Lelieveld et al., 2008; Kubistin et al., 2010). Here two tests are performed to evaluate the effect of isoprene oxidation and OH regeneration on model  $NO_2$  columns. The first test assumes a 100% regeneration of OH from the reaction of isoprene and OH (Case

- <sup>15</sup> 23 in Table 1) (Kubistin et al., 2010, assumed 130 %). This enhances the lower tropospheric (0–2 km) OH content by up to 40 % in July over the source regions of isoprene including parts of the southern provinces and the mountain and grassland areas in the north (not shown). Consequently, model NO<sub>2</sub> columns decrease by 0–4 % over most regions in July with smaller changes of within 2 % in January (Fig. 9a, b). Another test
- <sup>20</sup> is conducted by turning off the reaction of isoprene and OH to prevent the latter from being depleted (Case 24 in Table 1), following Kubistin et al. (2010). As a result, the lower tropospheric (0–2 km) OH content more than doubles in July over parts of the source regions of isoprene (not shown). Over the same regions, model NO<sub>2</sub> columns are increased mostly by 0–20 % with a drastic increase by 40–50 % in the southwest
- <sup>25</sup> (Fig. 9c). The increase of NO<sub>2</sub> is attributed to the suppressed formation of isoprene nitrates overcompensating for the effect of enhanced  $HNO_3$  production. The impact is smaller for January 2006 due primarily to lack of isoprene emissions (Fig. 9d).



Production of isoprene nitrates is an important sink of NO<sub>x</sub>. It is determined by the yield of isoprene nitrates and subsequent partial conversion back to NO<sub>x</sub> through photolysis or reactions with OH, i.e., the recycling of NO<sub>x</sub>. In addition, certain isoprene nitrates have relatively long atmospheric lifetime and can be transported to downwind areas before being transformed to NO<sub>x</sub> (Paulot et al., 2009). These factors affect the abundances of NO<sub>x</sub> in source and downwind regions with important consequences on the tropospheric chemistry. For example, previous model studies have suggested large sensitivity of modeled surface ozone concentration to assumptions on production and fate of isoprene nitrates (Fiore et al., 2005; Wu et al., 2007; Lin et al., 2008a,b,c;
10 Ito et al., 2009). The current estimates range from 2.4% to 15% for the net yield of isoprene nitrates (i.e., the amount that does not return to NO<sub>x</sub>) (Chen et al., 1998;

Sprengnether et al., 2002; Horowitz et al., 2003; Horowitz et al., 2007; Patchen et al., 2007; see summary by Paulot et al., 2009). A value of 10 %, on the high-end side, is adopted in GEOS-Chem by neglecting transport of the portion of isoprene nitrates de-<sup>15</sup> composed to NO<sub>x</sub>. Reducing the net yield to 2.4 %, a sensitivity simulation increases the NO<sub>2</sub> columns by up to 25 % over the southwest in July and by 0–6 % in January

(Case 25 in Table 1; Fig. 9e, f).

Production of peroxyacetyl nitrate (PAN) balanced by its thermal decomposition also affects the abundance of  $NO_2$ . Errors in its equilibrium constant are estimated to be about 20% at 298 K (Sander et al., 2011). A test simulation decreases the constant

about 20% at 298 K (Sander et al., 2011). A test simulation decreases the constant by 20%, resulting in slight enhancements of model NO<sub>2</sub> columns by 0–4% (within 2% over most regions) in both January and July (Case 26 in Table 1; figures are not shown).

#### 5.4 Heterogeneous reactions involving aerosols

<sup>25</sup> Uptake of HO<sub>2</sub> on aerosols reduces the HO<sub>x</sub> content in the troposphere and thus increases the NO<sub>x</sub> lifetime. The uptake coefficient depends on aerosol types and meteorological conditions, subject to large uncertainties (Jacob, 2000; Thornton et al., 2008; Kolb et al., 2010; Mao et al., 2010; Macintyre and Evans, 2011). It can be enhanced



by transition metal ions (TMIs) such as copper (Cu) and iron (Fe) contained in aqueous aerosols (Jacob, 2000; Thornton et al., 2008). Thornton et al. (2008) suggested an effective uptake coefficient of 0.04–0.1 on fine-mode pollution aerosols (radius ~ 100–200 nm) in the continental boundary layer, based on the mass fraction of Cu in near-surface aerosols observed over the contiguous USA. The average value of 0.07 is adopted in the current GEOS-Chem model for sulfate-nitrate-ammonium (SNA) aerosols with no regional dependence. The CTM does not take into account the effect of Fe, as considered to be much lower than that of Cu (Thornton et al., 2008). Meanwhile, the mass fraction of Cu measured in several cities in East China (2.9– 21.6 × 10<sup>-4</sup>) is higher than the average value over the USA (1.8 × 10<sup>-4</sup>) by a factor of 1.6–12 (He et al., 2001; Thornton et al., 2008; Yang et al., 2011). This is also supported by our preliminary measurement data in Beijing and Shanghai. In Beijing, the fraction of Cu doubled from 2000 to 2005 (He et al., 2001; Yang et al., 2011). Moreover, there is a large fraction of Fe in aerosols in Chinese cities (He et al., 2001; Yang et al., 2011)

- <sup>15</sup> with potentially important impacts on the uptake of HO<sub>2</sub>, due not just to its sole catalytic effect but also to its potential interactions with Cu enhancing the effects of both TMIs (Mao et al., 2012). Given the amount of TMIs, the uptake coefficient in these Chinese cities is likely to be much larger than currently assumed in the model. In a sensitivity simulation, the coefficient is increased to 0.2 in the troposphere over China (Case 27)
- in Table 1), a value consistent with previous estimates based on field measurements in North America and Hawaii (see review by Jacob, 2000). As a result, model NO<sub>2</sub> columns in July are enhanced by 4–14% over North China with smaller impacts in other areas (Fig. 10a). In January, model NO<sub>2</sub> columns increase by up to 50% in many areas (Fig. 10b). Another sensitivity simulation changes the uptake coefficient to 0.2
   on SNA aerosols in the PBL over China (Case 28 in Table 1). It thus increases the NO<sub>2</sub> columns by 2–10% over North China in July and by 2–16% over the south in January
  - (Fig. 10c, d).

Heterogeneous reaction of nitrogen pentoxide  $(N_2O_5)$  on aerosol surfaces is an important sink of  $NO_x$ , especially at night and in winter with relatively weak



photochemistry. The uptake coefficient of N<sub>2</sub>O<sub>5</sub> depends on aerosol types and meteorological conditions, as parameterized in the current GEOS-Chem (Evans and Jacob, 2005). More recent lab experiments suggested the coefficient to be lower than the Evans and Jacob (2005) estimate by a factor of up to 10 with dependence on meteorological conditions that is still not clear (Bertram et al., 2009; Brown et al., 2009). A sensitivity simulation reduces the uptake coefficient by an order of magnitude (Case 29 in Table 1). It consequently increases model NO<sub>2</sub> columns by 0–6 % over most regions in July and by more than 50 % over part of the northwest in January (Fig. 10e, f). The results are consistent with Lin and McElroy (2011).

#### **10 5.5 Aerosol optical properties**

The tropospheric photochemistry is affected by the amount of solar radiation scattered or absorbed by aerosols. Figures S14 and S15 compare modeled AOD with MODIS/Aqua measurements. In July, GEOS-Chem underestimates the MODIS AOD by 10–50 % in many areas of East China (Fig. S14). This is in part because the forma-

tion of secondary organic aerosols is not accounted for with the current model setup. For January, however, the CTM overestimates MODIS AOD by more than 100% in many places (Fig. S15). Further research is required to fully understand the causes of these differences.

Two test simulations are performed here to quantify the impacts of aerosol scattering and absorption on model NO<sub>2</sub> columns. The first test turns off both scattering and absorption of aerosols in calculating the photochemistry (Case 30 in Table 1). It thus increases model NO<sub>2</sub> columns in July by 0–6% over most regions but has a negative impact of 10–20% in January for a large portion of East China (Fig. 11a, b). Another test turns off the effect of black carbon alone resulting in reductions of NO<sub>2</sub> by 0–6%

<sup>25</sup> in July and by up to 20 % in January (Case 31 in Table 1; Fig. 11c, d). The differences (magnitude and sign) are evident between the two tests and between the two months. It is because most aerosols and  $NO_x$  are collocated in the PBL, in which situation scattering by aerosols alone tends to increase the actinic flux for photodissociation of



NO<sub>2</sub> while absorption by aerosols decreases the actinic flux. In July, the amount of black carbon (emitted mostly from the residential sector) is small relative to scattering aerosols, resulting in a dominant effect by scattering. In January, the concentration of black carbon increases; thus the aerosol scattering enhances the effect of absorption by providing more diffuse radiation, resulting in enhanced reduction of the actinic flux.

#### 5.6 Emissions

5

Emissions of CO and (to a lesser extent)  $SO_2$  affect the OH content with consequences on the reaction of OH and  $NO_2$  and other related photochemical processes. Large uncertainties, however, exist in current emission estimates based on both bottom-up and

- <sup>10</sup> top-down approaches (Fortems-Cheiney et al., 2011; Lu et al., 2011). Here a test simulation is performed to evaluate the impact of emissions of CO and SO<sub>2</sub> on model NO<sub>2</sub> columns, by increasing their emissions by 50 % (Case 32 in Table 1). It thus enhances the NO<sub>2</sub> columns by up to 6% over North China in July and by up to 10% over the south in January (Fig. S16a, b).
- Emissions of VOC affect OH and other radicals with consequences on the formation of HNO<sub>3</sub> and organic nitrates. Recent ground- and space-based measurements have suggested anthropogenic emissions of aromatics to be underestimated by a factor of 4–10 over East China in the INTEX-B inventory employed here (Liu et al., 2010, 2011). Meanwhile, the current model setup does not include the simulation of aro-
- <sup>20</sup> matics, preventing a direct evaluation of their impacts on model NO<sub>2</sub> columns. Analysis on propene and aromatics, however, suggests similarity in reactivity and oxidation products with respect to OH, pointing to a possibility of using propene as a proxy of aromatics in the model for purposes of sensitivity evaluation. Based on the reactivity and emission strength (provided from INTEX-B), it is estimated that assuming an extra
- $_{25}$  source (50 %) of propene can roughly represent the effect of aromatics on the nitrogen chemistry in accordance with the emission inventory. Therefore two tests are performed here to indirectly evaluate the impact of aromatics on model NO<sub>2</sub> columns, by increasing emissions of propene by 50 % and 300 % (i.e., assuming emissions of aromatics to



be six times as much as INTEX-B), respectively (Cases 33–34 in Table 1). The 50 % increase in propene emissions results in small changes (i.e., within 2 %) in model NO<sub>2</sub> columns for both months (Fig. S16c, d). Increasing emissions of propene by 300 % enhances the NO<sub>2</sub> columns by 0–4 % over most regions in July with a negative effect of

 $_5$  0–20% in most areas in January (Fig. 12). The different signs and magnitudes of NO<sub>2</sub> changes in the two months are attributed mainly to enhanced formation of PAN compensated by the effect of reduced production of HNO<sub>3</sub> through the NO<sub>2</sub> + OH reaction.

A further simulation increases emissions of CO,  $SO_2$  and VOC together by 50 %, following Lin et al. (2010b) (Case 35 in Table 1). As a result, model  $NO_2$  columns in July increase by up to 6 % over North China but decrease by up to 20 % over the southwest (Fig. S16a). The reduction in the southwest is attributed mainly to the increased forma-

(Fig. S16g). The reduction in the southwest is attributed mainly to the increased formation of organic nitrates. In January, changes in model  $NO_2$  columns are within 4 % over most regions (Fig. S16h).

Emissions of NO<sub>x</sub> assumed in GEOS-Chem may also affect the lifetime of NO<sub>x</sub> and <sup>15</sup> its partitioning into NO and NO<sub>2</sub> through the nonlinear chemical process. This effect has been estimated to be within 10 % for both January and July on the regional mean basis (Martin et al., 2006; Lin and McElroy, 2011; Lin, 2012).

#### 5.7 Other mechanisms and issues

Several new mechanisms have been proposed in recent years that may also affect the abundances of NO<sub>x</sub> in the troposphere. Li et al. (2008, 2009) suggested a reaction of NO<sub>2</sub>, excited by solar radiation > 420 nm, with water vapor to produce OH and nitrous acid (HONO). The reaction was estimated to be important at high solar zenith angles with low OH content, i.e., at twilight and/or at high latitudes. However, Carr et al. (2009) suggested a much smaller rate constant and thus negligible impacts on the tropospheric chemistry. The contradiction does not allow for conclusive determination of the true importance of the reaction (Sander et al., 2011). A recent study by Su et al. (2011) suggested a significant source of HONO from soil nitrites, particularly from fertilized



NO and OH, enhancing both source and sink of NO<sub>x</sub> likely with a small net impact on its abundance in the troposphere. Karl et al. (2010) suggested dry deposition of oxygenated VOC to be much larger than normally assumed over deciduous ecosystems. They suggested a consequent enhancement of OH by 0–4 % with a reduction of

<sup>5</sup> ozone by 1–3.5% over East China above the continental surface layer in September. The resulting impacts on model NO<sub>2</sub> columns are expected to be relatively small on the regional scale. Given the large uncertainties, these mechanisms are not included in most CTMs and are not evaluated quantitatively here.

Valin et al. (2011) suggested a potentially significant non-linear effect of model hori-

- <sup>10</sup> zontal resolution on simulations of NO<sub>2</sub> due to the interaction between NO<sub>x</sub> and HO<sub>x</sub>, based on chemical mechanism arguments and 7 days of WRF-Chem simulations (1–7 July, 2006). The effect is highly location and time dependent as affected by the varying meteorological and chemical conditions. It is expected to be smaller in winter (with lengthened lifetime and enhanced horizontal homogeneity of NO<sub>x</sub> loosening the re-
- quirement on model resolution) than in summer. Vertical resolution may also be important for nitrogen simulations considering the large vertical gradient of NO<sub>x</sub> and other related species (Lin and McElroy, 2010). Further research is needed to fully quantify the resolution-induced model errors.

## 6 Modifying model NO<sub>2</sub> columns accounting for errors in meteorology and chemistry

20

Model NO<sub>2</sub> columns can be modified on the first order by linearizing the impacts of errors in meteorological and chemical parameters:



$$\begin{aligned} \ln(\Omega) &= f(P_i), \qquad i = 1, 2, 3, \dots \\ &\approx \ln \Omega_0 + \sum \left( \partial \ln \Omega / \partial P_i \cdot \Delta P_i \right) \\ &\Rightarrow \Omega &= \Omega_0 \cdot \exp^{\sum \left( \partial \ln \Omega / \partial P_i \cdot \Delta P_i \right)} \end{aligned}$$

where Ω represents the post-modification NO<sub>2</sub> columns, Ω<sub>0</sub> represents modeled NO<sub>2</sub> columns prior to the modifications, *P<sub>i</sub>* represents a particular meteorological or chemical parameter, and *f*(*P<sub>i</sub>*) represents the CTM. ∂ ln Ω/∂*P<sub>i</sub>* is derived from the sensitivity simulations, and Δ*P<sub>i</sub>* is the quantified errors of the given parameter based on current knowledge. The modifications are done for gridboxes (at 0.25° long × 0.25° lat) with
 respect to the 284 ground meteorological stations, assuming that the impacts are local for NO<sub>2</sub> columns. To partially account for the effect of horizontal transport, the NO<sub>2</sub> columns are smoothed with values from the surrounding eight gridboxes (i.e., 3 gridboxes by 3 gridboxes smoothing) prior to the modifications.

Parameters considered for post-model modification include air temperature, water
vapor content, COD, uptake coefficients on aerosols for HO<sub>2</sub> and N<sub>2</sub>O<sub>5</sub>, rate constant for reaction of OH and NO<sub>2</sub>, OH regeneration from the isoprene chemistry, net yield of isoprene nitrates, aromatics (as represented by the 300 % increase in propene emissions), and scattering and absorption of radiation by aerosols (see Table 2). Potential errors in modeled PBL mixing are not accounted for here in lack of adequate measurements; the likely underestimate in the nighttime PBLH only has a small impact on modeled NO<sub>2</sub> columns in the afternoon (Fig. 6c, d; see Sect. 4.4). Errors in emissions of CO and SO<sub>2</sub> are also not taken into account due to lack of adequate information (Fortems-Cheiney et al., 2011; Lu et al., 2011). Errors in NO<sub>x</sub> emissions may affect modeling

of the non-linear photochemistry (see Sect. 5.6); they are however not accounted for <sup>25</sup> in this section aiming to evaluate the model performance with a given amount of NO<sub>x</sub> emissions.

As detailed in Table 2, the observed meteorological parameters are taken from the ISH and ISCCP datasets for parameter error calculations. The modeled effect of



(1)

aerosol scattering and absorption is adjusted based on the comparison between model and MODIS AOD. The best estimate rate constant for OH + NO<sub>2</sub> reaction is taken to be 15 % lower than adopted in the CTM (Mollner et al., 2010; Sander et al., 2011). The net yield of isoprene nitrates is best estimated at 6 % instead of 10 % (Paulot et al., 2009), neglecting for simplicity the horizontal transport of isoprene nitrates affecting

- the spatial distribution of  $NO_x$ . The uptake coefficient of  $N_2O_5$  on aerosols is taken to be 10% of the value adopted in GEOS-Chem (Bertram et al., 2009; Brown et al., 2009) as a highly simplified modification. A more appropriate value of 0.2 is assumed for the uptake coefficient of HO<sub>2</sub> to account for the large amount of TMIs in China (He
- et al., 2001; Thornton et al., 2008; Yang et al., 2011), as a rough estimate. In analyzing the uptake of  $HO_2$  and  $N_2O_5$ , the surface areas of aerosols are scaled by the ratio of MODIS AOD to model values. Note that these simplified modifications serve only to infer the impacts of potential errors in model meteorology and chemistry; and they cannot be interpreted literally as true apportionment of model errors.
- Table 2 and Figs. 13, 14 compare the NO<sub>2</sub> columns with and without post-model modifications. Figures 13, 14 also present results of the RMA regression for NO<sub>2</sub> columns pre- and post-modifications. Modification accounting for the uptake coefficient of HO<sub>2</sub> on aerosols has the largest impact on NO<sub>2</sub> columns across East China, resulting in a slope of about 1.16 in July and 1.10 in January (Figs. 13g, 14g). Modification for COD also has a large impact in July with a slope of about 1.12 (Fig. 13c). Medications
- for the rate constant of OH + NO<sub>2</sub> reaction and the approximate representation of aromatics using propene have important consequences in July and January, respectively (Figs. 13d, 14i).

Taking into account all modifications, the resulting NO<sub>2</sub> columns are highly correlated with the values prior to the modifications (Figs. 13a, 14a). They however are larger than model values in most places with a slope of about 1.39 in July and 1.09 for January under the RMA regression (Figs. 13a, 14a). Averaged over the stations, all modifications combined increase the NO<sub>2</sub> columns by 22 % in July and by 10 % in January. This suggests significant systematic biases in GEOS-Chem with important consequences



on the inverse modeling of  $NO_x$  emissions and other related applications. That model errors are likely larger in summer than in winter is in broad consistency with the findings of Huijnen et al. (2010) for Europe.

- van Noije et al. (2006) compared NO<sub>2</sub> columns for 2000 simulated by nine global
  <sup>5</sup> models differing in spatial resolution, meteorological inputs, soil/lightning emissions, uptake of N<sub>2</sub>O<sub>5</sub> on aerosols and other gaseous and heterogeneous chemical mechanisms. They found results for both July and January over Eastern China (110°–123° E, 30°–40° N) from three models to be different by up to 50 % from six other models providing more consistent simulations (including GEOS-Chem). van Noijie et al. (2006) also
  <sup>10</sup> found large inter-model spread in Europe, as confirmed by Huijnen et al. (2010) evalu-
- ating eight regional models and two global models. The inter-model differences in simulated NO<sub>2</sub> columns are likely because errors in meteorological/chemical parameters are inconsistent in sign and/or magnitude across the models. For example, parameters and processes may be overestimated by some models but neglected/underestimated
- <sup>15</sup> by other models.

20

25

Note that the modified columns are still lower than satellite retrievals with a slope of about 0.68 in July and 0.54 in January (Fig. 2c, d). The negative biases are driven mainly by those in polluted areas with retrieved NO<sub>2</sub> columns exceeding  $6 \times 10^{15}$  molecules cm<sup>-2</sup> in July and  $12 \times 10^{15}$  molecules cm<sup>-2</sup> in January. For cleaner areas, the slope increases to about 1.06 in July and 0.91 in January.

#### 7 Implications for surface ozone simulation of errors in model meteorological and chemical parameters

Errors in model meteorology and chemistry also have significant impacts on simulations of surface ozone. Here the post-model modification (as done for NO<sub>2</sub> in Sect. 6) is implemented to simulated ozone concentrations to evaluate the respective model performance. Results are presented in Table 2 and Figs. 15, 16. They should be interpreted with caution since the effect of horizontal transport is not fully taken into account



by the localized station-by-station adjustments, in addition to the linearization assumption and simplified error quantification for model parameters.

Modification accounting for the uptake coefficient of HO<sub>2</sub> on aerosols has the largest impact on surface ozone. In July, it results in a slope of 0.93 with a  $R^2$  of 0.89 from the

- RMA regression (Fig. 15g). More importantly, it reduces the peak ozone concentration by about 15 ppb. The large sensitivity of  $O_3$  formation to HO<sub>2</sub> uptake efficiency was also found by recent observation-based modeling studies over the region (Kanaya, et al., 2009; Liu et al., 2012). In January, the impacts are more location-dependent with the  $R^2$ decreasing to about 0.71 (Fig. 16g). Modifications for COD and aromatics (represented
- by propene) also have significant consequences on surface ozone in January, followed by modifications for AOD and water vapor (Fig. 16b, c, i, j). For July, modifications are important on temperature, water vapor, COD and aromatics (Fig. 15a, b, c, i).

All modifications combined result in a slope of about 0.86 with an intercept of 7.4 ppb in July for surface ozone (Fig. 15a). Spatially, ozone concentrations are reduced in the

- north but enhanced in the south resulting in a  $R^2$  of about 0.82 (Fig. 15a). For January, 15 ozone concentrations are decreased in the west but increased in the east resulting in a weak spatial correlation with a  $R^2$  of 0.22 only (Fig. 16a). The spatial pattern is attributed primarily to the negative impacts of modifications for COD and HO<sub>2</sub> uptake coefficient counteracted by the positive effect of inclusion of aromatics (represented by increased propene emissions).
- 20

#### Conclusions 8

25

Satellite remote sensing has been proven useful in constraining emissions of NO<sub>v</sub>, particularly for China and other developing countries for which bottom-up inventories contain large uncertainties and/or are not updated timely to reflect the rapid emission growth. Such top-down constraint is subject to errors not just in satellite retrievals but also in model simulations interpreting the relation between observed abundance of nitrogen and its emissions. Quantifying the model errors over China is however



challenging due to lack of sufficient in-situ measurements of nitrogen. In this study, an alternate approach is proposed for model evaluation over East China by conducting a systematic analysis of meteorological and chemical parameters affecting the simulation of  $NO_x$ . The analysis is focused in January and July 2006 to evaluate the seasonal dependence of model sensitivity to meteorology and chemistry.

As a demonstration, the nested version for Asia of the widely used GEOS-Chem model is evaluated here. The CTM is driven by the GEOS-5 meteorological fields and the INTEX-B anthropogenic emissions for Asia, as used in the inverse modeling of NO<sub>x</sub> emissions and many other applications (Lin et al., 2010a,b; Lin and McElroy, 2010, 2011; Lin, 2012; Wang et al., 2012) (see also http://acmg.seas.harvard.edu/ recentpapers.html). The nested model has been found to underestimate NO<sub>2</sub> columns retrieved from OMI (DOMINO-2) by about 20% in July and by as large as 36% in January averaged over East China (Lin, 2012). Modeled NO<sub>2</sub> columns are only about 20–30% of retrieved values over extremely polluted areas such as cities and those with

<sup>15</sup> large point sources, particular in the north in January (Lin, 2012).

5

Errors in the GEOS-5 meteorological fields are evaluated with observations from the ground network (NCDC ISH) and from space (ISCCP). GEOS-5 significantly underestimates cloud fraction and cloud optical depth over East China as compared to measurements from ISCCP. It also contains errors of regional and seasonal dependence

in air temperature, relative humidity, water vapor content, wind speed and precipitation. Sensitivity simulations suggest that the impacts of these errors on modeled NO<sub>2</sub> columns are normally below 20 % across East China.

Errors in PBL mixing affecting the vertical distribution of NO<sub>2</sub> are difficult to quantify due to lack of direct observations. A series of sensitivity simulations perturbing the

PBLH show significant influences on model NO<sub>2</sub> columns (when the averaging kernel from the satellite product is applied to simulated vertical distribution of NO<sub>2</sub>). A 100 % increase in PBLH results in closer agreement between modeled and retrieved columns averaged over East China; it however cannot eliminate the negative biases in extremely polluted areas.



A variety of chemical parameters are also evaluated with the CTM based on current understanding of their uncertainties. Factors considered include the rate constant for reaction of OH + NO<sub>2</sub>, yield of HNO<sub>3</sub> from reaction of NO + HO<sub>2</sub>, OH regeneration from the isoprene chemistry, yield and fate of isoprene nitrates, equilibrium constant for formation and thermal decomposition of PAN, uptake coefficients on aerosols for HO<sub>2</sub> and N<sub>2</sub>O<sub>5</sub>, chemistry of aromatics, emissions of CO, VOC and SO<sub>2</sub>, and scattering and absorption of solar radiation by aerosols. Perturbations on these factors affect simulated NO<sub>2</sub> columns by various magnitudes strongly depending on regions and seasons.

- Attempts are made to modify the columns simulated at the 284 ground meteorological stations in East China, accounting for errors in various meteorological and chemical parameters. The modifications together increase the NO<sub>2</sub> columns by 22 % in July and by 10 % in January averaged across the stations, likely indicating systematic model errors. They result in closer agreement with satellite retrievals, but cannot eliminate the
- <sup>15</sup> negative biases in extremely polluted places. Individually, assuming the uptake coefficient of HO<sub>2</sub> on aerosols to be 0.2 (to account for the large amount of transition metal ions) has the largest impact on NO<sub>2</sub> columns, followed by corrections on COD, rate constant of OH + NO<sub>2</sub> reaction, uptake coefficient of N<sub>2</sub>O<sub>5</sub> on aerosols, and inclusion of aromatics.
- Errors in meteorological and chemical parameters also have significant consequences on simulations of ozone and its regional and global transport. A similar post-model modification for surface ozone in July results in decreases of ozone concentrations in the north with increases in the south, both by about 0–20%. In particular, the peak values are reduced by about 15 ppb. For January, ozone concentrations are reduced in the west and enhanced in the east, with the differences exceeding 20% in
- reduced in the west and enhanced in the east, with the differences exceeding 20% in many places.

It is important to understand the large differences between modeled and retrieved tropospheric  $NO_2$  columns in extremely polluted areas. Model errors appear not to be the primary cause based on the present analysis. Satellite retrievals are sensitive



to assumptions on the amount of stratospheric NO<sub>2</sub>, aerosols, clouds, surface reflectance, terrain height, and a priori vertical profile of NO<sub>2</sub>. Most aerosols of anthropogenic origin are collocated with NO<sub>2</sub> in the PBL, affecting the amount of solar radiation absorbed by NO<sub>2</sub>. The presence of aerosols is not accounted for explicitly in current satellite NO<sub>2</sub> products; and their scattering effect is treated implicitly with arbitrarily enhanced cloud fraction. The sensitivity of NO<sub>2</sub> columns on aerosol scattering and absorption is found in this study to be within 20 % across East China, a value much smaller than the magnitude of model underestimation in extremely polluted areas. The presence of aerosols may also affect the retrieval of surface reflectance with consequences on subsequent cloud and NO<sub>2</sub> retrievals. This factor cannot likely account for the large model underestimation in extremely polluted areas. In addition, errors (in percentage) in satellite retrievals caused by non-aerosol factors are not expected to be highly correlated to the amount of NO<sub>2</sub>. Therefore retrieval errors seem unlikely to be the main source of model-retrieval differences in extremely polluted areas. It appears

that the differences are derived more likely from the underestimate of  $NO_x$  emissions in the INTEX-B inventory used to drive the model simulations.

### Supplementary material related to this article is available online at: http://www.atmos-chem-phys-discuss.net/12/14269/2012/ acpd-12-14269-2012-supplement.pdf.

Acknowledgements. This research is supported by the National Natural Science Foundation of China, grant 41005078 and 41175127. The work of G. Zhuang is supported by the great international collaboration project of MOST, China (2010DFA92230), National Natural Science Foundation of China (Grant No. 41128005, fund for collaboration with oversea scholars). We acknowledge the free use of tropospheric NO<sub>2</sub> column data from www.temis.nl, MODIS AOD data from NASA, and meteorological data from NOAA NCDC and the ISCCP project. We thank Randall V. Martin and Mathew Evans for useful comments.



#### References

15

- van der A, R. J., Peters, D., Eskes, H., Boersma, K. F., Van Roozendael, M., De Smedt, I., and Kelder, H. M.: Detection of the trend and seasonal variation in tropospheric NO<sub>2</sub> over China, J. Geophys. Res.-Atmos., 111, D12317, doi:10.1029/2005jd006594, 2006.
- <sup>5</sup> Atkinson, R., Baulch, D. L., Cox, R. A., Crowley, J. N., Hampson, R. F., Hynes, R. G., Jenkin, M. E., Rossi, M. J., and Troe, J.: Evaluated kinetic and photochemical data for atmospheric chemistry: Volume I – gas phase reactions of O<sub>x</sub>, HO<sub>x</sub>, NO<sub>x</sub> and SO<sub>x</sub> species, Atmos. Chem. Phys., 4, 1461–1738, doi:10.5194/acp-4-1461-2004, 2004.

Atkinson, R., Baulch, D. L., Cox, R. A., Crowley, J. N., Hampson, R. F., Hynes, R. G.,

- Jenkin, M. E., Rossi, M. J., Troe, J., and IUPAC Subcommittee: Evaluated kinetic and photochemical data for atmospheric chemistry: Volume II – gas phase reactions of organic species, Atmos. Chem. Phys., 6, 3625–4055, doi:10.5194/acp-6-3625-2006, 2006.
  - Bertram, T. H., Thornton, J. A., Riedel, T. P., Middlebrook, A. M., Bahreini, R., Bates, T. S., Quinn, P. K., and Coffman, D. J.: Direct observations of N<sub>2</sub>O<sub>5</sub> reactivity on ambient aerosol particles, Geophys. Res. Lett., 36, L19803, doi:10.1029/2009gl040248, 2009.
- Boersma, K. F., Jacob, D. J., Trainic, M., Rudich, Y., DeSmedt, I., Dirksen, R., and Eskes, H. J.: Validation of urban NO<sub>2</sub> concentrations and their diurnal and seasonal variations observed from the SCIAMACHY and OMI sensors using in situ surface measurements in Israeli cities, Atmos. Chem. Phys., 9, 3867–3879, doi:10.5194/acp-9-3867-2009, 2009.
- Boersma, K. F., Jacob, D. J., Bucsela, E. J., Perring, A. E., Dirksen, R., van der A, R. J., Yantosca, R. M., Park, R. J., Wenig, M. O., Bertram, T. H., and Cohen, R. C.: Validation of OMI tropospheric NO<sub>2</sub> observations during INTEX-B and application to constrain NO<sub>x</sub> emissions over the Eastern United States and Mexico, Atmos. Environ., 42, 4480–4497, doi:10.1016/j.atmosenv.2008.02.004, 2008.
- <sup>25</sup> Boersma, K. F., Eskes, H. J., Dirksen, R. J., van der A, R. J., Veefkind, J. P., Stammes, P., Huijnen, V., Kleipool, Q. L., Sneep, M., Claas, J., Leitão, J., Richter, A., Zhou, Y., and Brunner, D.: An improved tropospheric NO<sub>2</sub> column retrieval algorithm for the Ozone Monitoring Instrument, Atmos. Meas. Tech., 4, 1905–1928, doi:10.5194/amt-4-1905-2011, 2011.

Brinksma, E. J., Pinardi, G., Volten, H., Braak, R., Richter, A., Schonhardt, A., van Roozendael, M., Fayt, C., Hermans, C., Dirksen, R. J., Vlemmix, T., Berkhout, A. J. C., Swart, D. P. J.,
Oetjen, H., Wittrock, F., Wagner, T., Ibrahim, O. W., de Leeuw, G., Moerman, M.,
Curier, R. L., Celarier, E. A., Cede, A., Knap, W. H., Veefkind, J. P., Eskes, H. J., Al-



laart, M., Rothe, R., Piters, A. J. M., and Levelt, P. F.: The 2005 and 2006 DANDELIONS  $NO_2$  and aerosol intercomparison campaigns, J. Geophys. Res.-Atmos., 113, D16S46, doi:10.1029/2007jd008808, 2008.

Brown, S. S., Dube, W. P., Fuchs, H., Ryerson, T. B., Wollny, A. G., Brock, C. A., Bahreini, R.,

- Middlebrook, A. M., Neuman, J. A., Atlas, E., Roberts, J. M., Osthoff, H. D., Trainer, M., Fehsenfeld, F. C., and Ravishankara, A. R.: Reactive uptake coefficients for N<sub>2</sub>O<sub>5</sub> determined from aircraft measurements during the Second Texas Air Quality Study: comparison to current model parameterizations, J. Geophys. Res.-Atmos., 114, D00F10, doi:10.1029/2008jd011679, 2009.
- <sup>10</sup> Bucsela, E. J., Perring, A. E., Cohen, R. C., Boersma, K. F., Celarier, E. A., Gleason, J. F., Wenig, M. O., Bertram, T. H., Wooldridge, P. J., Dirksen, R., and Veefkind, J. P.: Comparison of tropospheric NO<sub>2</sub> from in situ aircraft measurements with near-real-time and standard product data from OMI, J. Geophys. Res.-Atmos., 113, D16S31, doi:10.1029/2007jd008838, 2008.
- <sup>15</sup> Butkovskaya, N., Kukui, A., and Le Bras, G.: HNO<sub>3</sub> forming channel of the HO<sub>2</sub> + NO reaction as a function of pressure and temperature in the ranges of 72–600 torr and 223–323 K, J. Phys. Chem. A, 111, 9047–9053, doi:10.1021/jp074117m, 2007.
  - Butkovskaya, N., Rayez, M.-T., Rayez, J.-C., Kukui, A., and Le Bras, G.: Water vapor effect on the HNO<sub>3</sub> yield in the HO<sub>2</sub> + NO reaction: experimental and theoretical evidence, J. Phys. Chem. A, 113, 11327–11342, doi:10.1021/jp811428p, 2009.

20

30

Butkovskaya, N. I., Kukui, A., Pouvesle, N., and Le Bras, G.: Formation of nitric acid in the gasphase HO<sub>2</sub> + NO reaction: effects of temperature and water vapor, J. Phys. Chem. A, 109, 6509–6520, doi:10.1021/jp051534v, 2005.

Butler, T. M., Taraborrelli, D., Brühl, C., Fischer, H., Harder, H., Martinez, M., Williams, J.,

- Lawrence, M. G., and Lelieveld, J.: Improved simulation of isoprene oxidation chemistry with the ECHAM5/MESSy chemistry-climate model: lessons from the GABRIEL airborne field campaign, Atmos. Chem. Phys., 8, 4529–4546, doi:10.5194/acp-8-4529-2008, 2008.
  - Carr, S., Heard, D. E., and Blitz, M. A.: Comment on "Atmospheric hydroxyl radical production from electronically excited NO<sub>2</sub> and H<sub>2</sub>O", Science, 324, p. 336, doi:10.1126/science.1166669, 2009.
  - Chen, D., Wang, Y., McElroy, M. B., He, K., Yantosca, R. M., and Le Sager, P.: Regional CO pollution and export in China simulated by the high-resolution nested-grid GEOS-Chem model, Atmos. Chem. Phys., 9, 3825–3839, doi:10.5194/acp-9-3825-2009, 2009.



- Chen, X. H., Hulbert, D., and Shepson, P. B.: Measurement of the organic nitrate yield from OH reaction with isoprene, J. Geophys. Res.-Atmos., 103, 25563–25568, doi:10.1029/98jd01483, 1998.
- Emmerson, K. M. and Evans, M. J.: Comparison of tropospheric gas-phase chemistry schemes
- <sup>5</sup> for use within global models, Atmos. Chem. Phys., 9, 1831–1845, doi:10.5194/acp-9-1831-2009, 2009.
  - Evans, M. J. and Jacob, D. J.: Impact of new laboratory studies of N<sub>2</sub>O<sub>5</sub> hydrolysis on global model budgets of tropospheric nitrogen oxides, ozone, and OH, Geophys. Res. Lett., 32, L09813, doi:10.1029/2005gl022469, 2005.
- Fiore, A. M., Horowitz, L. W., Purves, D. W., Levy, H., Evans, M. J., Wang, Y. X., Li, Q. B., and Yantosca, R. M.: Evaluating the contribution of changes in isoprene emissions to surface ozone trends over the Eastern United States, J. Geophys. Res.-Atmos., 110, D12303, doi:10.1029/2004jd005485, 2005.

Fortems-Cheiney, A., Chevallier, F., Pison, I., Bousquet, P., Szopa, S., Deeter, M. N., and Cler-

- baux, C.: Ten years of CO emissions as seen from Measurements of Pollution in the Troposphere (MOPITT), J. Geophys. Res.-Atmos., 116, D05304, doi:10.1029/2010jd014416, 2011.
  - Global Modeling and Assimilation Office: File Specification for GEOS-5 DAS Gridded Output Document No. GMAO-1001v6.1, Global Modeling and Assimilation Office, Earth Sciences Division NASA Goddard Space Flight Center, Greenbelt, Maryland 20771, 2006.
- Division NASA Goddard Space Flight Center, Greenbelt, Maryland 20771, 2006.
   He, K. B., Yang, F. M., Ma, Y. L., Zhang, Q., Yao, X. H., Chan, C. K., Cadle, S., Chan, T., and Mulawa, P.: The characteristics of PM<sub>2.5</sub> in Beijing, China, Atmos. Environ., 35, 4959–4970, 2001.

Holtslag, A. and Boville, B.: Local versus nonlocal boundary-layer diffusion in a global climate model, J. Climate, 6, 1825–1842, 1993.

Horowitz, L. W., Fiore, A. M., Milly, G. P., Cohen, R. C., Perring, A., Wooldridge, P. J., Hess, P. G., Emmons, L. K., and Lamarque, J.-F.: Observational constraints on the chemistry of isoprene nitrates over the Eastern United States, J. Geophys. Res.-Atmos., 112, D12S08, doi:10.1029/2006jd007747, 2007.

25

<sup>30</sup> Horowitz, L. W., Walters, S., Mauzerall, D. L., Emmons, L. K., Rasch, P. J., Granier, C., Tie, X. X., Lamarque, J. F., Schultz, M. G., Tyndall, G. S., Orlando, J. J., and Brasseur, G. P.: A global simulation of tropospheric ozone and related tracers: description and evaluation of MOZART, version 2, J. Geophys. Res.-Atmos., 108, 4784–4508, doi:10.1029/2002jd002853, 2003.



Hudman, R. C., Jacob, D. J., Turquety, S., Leibensperger, E. M., Murray, L. T., Wu, S., Gilliland, A. B., Avery, M., Bertram, T. H., Brune, W., Cohen, R. C., Dibb, J. E., Flocke, F. M., Fried, A., Holloway, J., Neuman, J. A., Orville, R., Perring, A., Ren, X., Sachse, G. W., Singh, H. B., Swanson, A., and Wooldridge, P. J.: Surface and lightning sources of nitrogen oxides over the United States: magnitudes, chemical evolution, and outflow, J. Geophys. Res.-Atmos., 112, D12S05, doi:10.1029/2006jd007912, 2007.

5

20

- Huijnen, V., Eskes, H. J., Poupkou, A., Elbern, H., Boersma, K. F., Foret, G., Sofiev, M., Valdebenito, A., Flemming, J., Stein, O., Gross, A., Robertson, L., D'Isidoro, M., Kioutsioukis, I., Friese, E., Amstrup, B., Bergstrom, R., Strunk, A., Vira, J., Zyryanov, D., Mau-
- <sup>10</sup> rizi, A., Melas, D., Peuch, V.-H., and Zerefos, C.: Comparison of OMI NO<sub>2</sub> tropospheric columns with an ensemble of global and European regional air quality models, Atmos. Chem. Phys., 10, 3273–3296, doi:10.5194/acp-10-3273-2010, 2010.
  - Ito, A., Sillman, S., and Penner, J. E.: Global chemical transport model study of ozone response to changes in chemical kinetics and biogenic volatile organic compounds emissions due to
- <sup>15</sup> increasing temperatures: sensitivities to isoprene nitrate chemistry and grid resolution, J. Geophys. Res.-Atmos., 114, D09301, doi:10.1029/2008jd011254, 2009.
  - Jacob, D. J.: Heterogeneous chemistry and tropospheric ozone, Atmos. Environ., 34, 2131–2159, doi:10.1016/s1352-2310(99)00462-8, 2000.

Jaeglé, L., Steinberger, L., Martin, R. V., and Chance, K.: Global partitioning of NO<sub>x</sub> sources using satellite observations: Relative roles of fossil fuel combustion, biomass burning and

- soil emissions, Faraday Discuss., 130, 407–423, doi:10.1039/b502128f, 2005. Kanaya, Y., Pochanart, P., Liu, Y., Li, J., Tanimoto, H., Kato, S., Suthawaree, J., Inomata, S., Taketani, F., Okuzawa, K., Kawamura, K., Akimoto, H., and Wang, Z. F.: Rates and regimes of photochemical ozone production over Central East China in June 2006: a box model
- analysis using comprehensive measurements of ozone precursors, Atmos. Chem. Phys., 9, 7711–7723, doi:10.5194/acp-9-7711-2009, 2009.
  - Karl, T., Harley, P., Emmons, L., Thornton, B., Guenther, A., Basu, C., Turnipseed, A., and Jardine, K.: Efficient Atmospheric Cleansing of Oxidized Organic Trace Gases by Vegetation, Science, 330, 816–819, doi:10.1126/science.1192534, 2010.
- <sup>30</sup> Kolb, C. E., Cox, R. A., Abbatt, J. P. D., Ammann, M., Davis, E. J., Donaldson, D. J., Garrett, B. C., George, C., Griffiths, P. T., Hanson, D. R., Kulmala, M., McFiggans, G., Pöschl, U., Riipinen, I., Rossi, M. J., Rudich, Y., Wagner, P. E., Winkler, P. M., Worsnop, D. R., and O' Dowd, C. D.: An overview of current issues in the uptake of atmospheric trace gases by



aerosols and clouds, Atmos. Chem. Phys., 10, 10561-10605, doi:10.5194/acp-10-10561-2010, 2010.

- Kubistin, D., Harder, H., Martinez, M., Rudolf, M., Sander, R., Bozem, H., Eerdekens, G., Fischer, H., Gurk, C., Klüpfel, T., Königstedt, R., Parchatka, U., Schiller, C. L., Stickler, A.,
- <sup>5</sup> Taraborrelli, D., Williams, J., and Lelieveld, J.: Hydroxyl radicals in the tropical troposphere over the Suriname rainforest: comparison of measurements with the box model MECCA, Atmos. Chem. Phys., 10, 9705–9728, doi:10.5194/acp-10-9705-2010, 2010.

Lamsal, L. N., Martin, R. V., van Donkelaar, A., Celarier, E. A., Bucsela, E. J., Boersma, K. F., Dirksen, R., Luo, C., and Wang, Y.: Indirect validation of tropospheric nitrogen diox-

- <sup>10</sup> ide retrieved from the OMI satellite instrument: insight into the seasonal variation of nitrogen oxides at northern midlatitudes, J. Geophys. Res.-Atmos., 115, D05302, doi:10.1029/2009jd013351, 2010.
  - Lamsal, L. N., Martin, R. V., Padmanabhan, A., van Donkelaar, A., Zhang, Q., Sioris, C. E., Chance, K., Kurosu, T. P., and Newchurch, M. J.: Application of satellite observations for
- timely updates to global anthropogenic  $NO_x$  emission inventories, Geophys. Res. Lett., 38, L05810, doi:10.1029/2010gl046476, 2011.
  - Lelieveld, J., Butler, T. M., Crowley, J. N., Dillon, T. J., Fischer, H., Ganzeveld, L., Harder, H., Lawrence, M. G., Martinez, M., Taraborrelli, D., and Williams, J.: Atmospheric oxidation capacity sustained by a tropical forest, Nature, 452, 737–740, doi:10.1038/nature06870, 2008.
- <sup>20</sup> Li, S., Matthews, J., and Sinha, A.: Atmospheric hydroxyl radical production from electronically excited NO<sub>2</sub> and H<sub>2</sub>O, Science, 319, 1657–1660, doi:10.1126/science.1151443, 2008.
  - Li, S., Matthews, J., and Sinha, A.: Response to Comment on "Atmospheric hydroxyl radical production from electronically excited  $NO_2$  and  $H_2O$ ", Science, 324, p. 336, doi:10.1126/science.1166877, 2009.
- Lin, J.-T.: Satellite constraint for emissions of nitrogen oxides from anthropogenic, lightning and soil sources over East China on a high-resolution grid, Atmos. Chem. Phys., 12, 2881–2898, doi:10.5194/acp-12-2881-2012, 2012.
  - Lin, J. T. and McElroy, M. B.: Impacts of boundary layer mixing on pollutant vertical profiles in the lower troposphere: implications to satellite remote sensing, Atmos. Environ., 44, 1726–
- <sup>30</sup> 1739, doi:10.1016/j.atmosenv.2010.02.009, 2010.
  - Lin, J.-T. and McElroy, M. B.: Detection from space of a reduction in anthropogenic emissions of nitrogen oxides during the Chinese economic downturn, Atmos. Chem. Phys., 11, 8171–8188, doi:10.5194/acp-11-8171-2011, 2011.



Lin, J.-T., Liang, X.-Z., and Wuebbles, D. J.: Effects of intercontinental transport on surface ozone over the United States: present and future assessment with a global model, Geophys. Res. Lett., 35, L02805, doi:10.1029/2007GL031415, 2008a.

Lin, J.-T., Youn, D., Liang, X.-Z., and Wuebbles, D. J.: Global model simulation of summertime US ozone diurnal cycle and its sensitivity to PBL mixing, spatial resolution, and emissions,

- 5 US ozone diurnal cycle and its sensitivity to PBL mixing, spatial resolution, and emissions Atmos. Environ., 42, 8470–8483, doi:10.1016/j.atmosenv.2008.08.012, 2008b.
  - Lin, J.-T., Patten, K. O., Hayhoe, K., Liang, X.-Z., and Wuebbles, D. J.: Effects of future climate and biogenic emission changes on surface ozone over the United States and China, J. Appl. Meteor., 47, 1888–1909, doi:10.1175/2007JAMC1681.1, 2008c.
- Lin, J.-T., Nielsen, C. P., Zhao, Y., Lei, Y., Liu, Y., and McElroy, M. B.: Recent changes in particulate air pollution over China observed from space and the ground: effectiveness of emission control, Environ. Sci. Technol., 44, 7771–7776, doi:10.1021/es101094t, 2010a.
  - Lin, J.-T., McElroy, M. B., and Boersma, K. F.: Constraint of anthropogenic NO<sub>x</sub> emissions in China from different sectors: a new methodology using multiple satellite retrievals, Atmos. Chem. Phys., 10, 63–78, doi:10.5194/acp-10-63-2010, 2010b.
- Liu, H., Crawford, J. H., Considine, D. B., Platnick, S., Norris, P. M., Duncan, B. N., Pierce, R. B., Gao, C., and Yantosca, R. M.: Sensitivity of photolysis frequencies and key tropospheric oxidants in a global model to cloud vertical distributions and optical properties, J. Geophys. Res., 114, D10305, doi:10.1029/2008jd011503, 2009.

15

25

- Liu, S. and Liang, X.-Z.: Observed diurnal cycle climatology of planetary boundary layer height, J. Climate, 23, 5790–5809, doi:10.1175/2010jcli3552.1, 2010.
  - Liu, Z., Wang, Y., Zhao, C., Vrekoussis, M., Richter, A., Wittrock, F., Burrow, J. P., Shao, M., Liu, S.-C., Chang, C.-C., Wang, H., and Chen, C.: The missing source of glyoxal over China and its implications on organic aerosol budgets, AGU Fall Meeting 2011, San Francisco, abstract #A21I-07, 2011.
  - Liu, Z., Wang, Y., Gu, D., Zhao, C., Huey, L. G., Stickel, R., Liao, J., Shao, M., Zhu, T., Zeng, L., Liu, S.-C., Chang, C.-C., Amoroso, A., and Costabile, F.: Evidence of reactive aromatics as a major source of peroxy acetyl nitrate over China, Environ. Sci. Technol., 44, 7017–7022, doi:10.1021/es1007966, 2010.
- Liu, Z., Wang, Y., Gu, D., Zhao, C., Huey, L. G., Stickel, R., Liao, J., Shao, M., Zhu, T., Zeng, L., Amoroso, A., Costabile, F., Chang, C.-C., and Liu, S.-C.: Summertime photochemistry during CAREBeijing-2007: RO<sub>x</sub> budgets and O<sub>3</sub> formation, Atmos. Chem. Phys. Discuss., 12, 4679– 4717, doi:10.5194/acpd-12-4679-2012, 2012.





- Lu, Z., Zhang, Q., and Streets, D. G.: Sulfur dioxide and primary carbonaceous aerosol emissions in China and India, 1996–2010, Atmos. Chem. Phys., 11, 9839–9864, doi:10.5194/acp-11-9839-2011, 2011.
- Macintyre, H. L. and Evans, M. J.: Parameterisation and impact of aerosol uptake of  $HO_2$  on
- <sup>5</sup> a global tropospheric model, Atmos. Chem. Phys., 11, 10965–10974, doi:10.5194/acp-11-10965-2011, 2011.
  - Mao, J., Jacob, D. J., Evans, M. J., Olson, J. R., Ren, X., Brune, W. H., Clair, J. M. St., Crounse, J. D., Spencer, K. M., Beaver, M. R., Wennberg, P. O., Cubison, M. J., Jimenez, J. L., Fried, A., Weibring, P., Walega, J. G., Hall, S. R., Weinheimer, A. J., Co-
- <sup>10</sup> hen, R. C., Chen, G., Crawford, J. H., McNaughton, C., Clarke, A. D., Jaeglé, L., Fisher, J. A., Yantosca, R. M., Le Sager, P., and Carouge, C.: Chemistry of hydrogen oxide radicals (HO<sub>x</sub>) in the Arctic troposphere in spring, Atmos. Chem. Phys., 10, 5823–5838, doi:10.5194/acp-10-5823-2010, 2010.

Martin, R. V., Sioris, C. E., Chance, K., Ryerson, T. B., Bertram, T. H., Wooldridge, P. J.,

<sup>15</sup> Cohen, R. C., J. Andy Neuman, Swanson, A., and Flocke, F. M.: Evaluation of spacebased constraints on global nitrogen oxide emissions with regional aircraft measurements over and downwind of Eastern North America, J. Geophys. Res., 111, D15308, doi:10.1029/2005JD006680, 2006.

Mijling, B., van der A, R. J., Boersma, K. F., Van Roozendael, M., De Smedt, I., and

- <sup>20</sup> Kelder, H. M.: Reductions of NO<sub>2</sub> detected from space during the 2008 Beijing Olympic Games, Geophys. Res. Lett., 36, L13801, doi:10.1029/2009gl038943, 2009.
  - Mollner, A. K., Valluvadasan, S., Feng, L., Sprague, M. K., Okumura, M., Milligan, D. B., Bloss, W. J., Sander, S. P., Martien, P. T., Harley, R. A., McCoy, A. B., and Carter, W. P. L.: Rate of gas phase association of hydroxyl radical and nitrogen dioxide, Science, 330, 646– 649, doi:10.1126/science.1193030, 2010.
  - Mao, J., Fan S., and Jacob, D. J.: Radical loss in the atmosphere from Cu-Fe redox coupling in aerosols, Nature, in review, 2012.

25

- van Noije, T. P. C., Eskes, H. J., Dentener, F. J., Stevenson, D. S., Ellingsen, K., Schultz, M. G., Wild, O., Amann, M., Atherton, C. S., Bergmann, D. J., Bey, I., Boersma, K. F., Butler, T.,
- <sup>30</sup> Cofala, J., Drevet, J., Fiore, A. M., Gauss, M., Hauglustaine, D. A., Horowitz, L. W., Isaksen, I. S. A., Krol, M. C., Lamarque, J.-F., Lawrence, M. G., Martin, R. V., Montanaro, V., Müller, J.-F., Pitari, G., Prather, M. J., Pyle, J. A., Richter, A., Rodriguez, J. M., Savage, N. H., Strahan, S. E., Sudo, K., Szopa, S., and van Roozendael, M.: Multi-model ensemble simula-



tions of tropospheric  $NO_2$  compared with GOME retrievals for the year 2000, Atmos. Chem. Phys., 6, 2943–2979, doi:10.5194/acp-6-2943-2006, 2006.

- Okumura, M. and Sander, S. P.: Gas-Phase Formation Rates of Nitric Acid and its Isomers under Urban Conditions, State of California Air Resources Board, Sacramento, California
   95814, 2005.
  - Patchen, A. K., Pennino, M. J., Kiep, A. C., and Elrod, M. J.: Direct kinetics study of the productforming channels of the reaction of isoprene-derived hydroxyperoxy radicals with NO, Int. J. Chem. Kinet., 39, 353–361, doi:10.1002/kin.20248, 2007.

Paulot, F., Crounse, J. D., Kjaergaard, H. G., Kroll, J. H., Seinfeld, J. H., and Wennberg, P. O.:

<sup>10</sup> Isoprene photooxidation: new insights into the production of acids and organic nitrates, Atmos. Chem. Phys., 9, 1479–1501, doi:10.5194/acp-9-1479-2009, 2009.

Pryor, S. C. and Barthelmie, R. J.: Assessing climate change impacts on the near-term stability of the wind energy resource over the United States, Proc. Natl. Acad. Sci., 108, 8167–8171, doi:10.1073/pnas.1019388108, 2011.

Richter, A., Burrows, J. P., Nüß, H., Granier, C., and Niemeier, U.: Increase in tropospheric nitrogen dioxide over China observed from space, Nature, 437, 129–132, doi:10.1038/nature04092, 2005.

Sander, S. P., Abbatt, J. P. D., Barker, J. R., Burkholder, J. B., Friedl, R. R., Golden, D. M., Huie, R. E., Kolb, C. E., Kurylo, M. J., Moortgat, G. K., Orkin, V. L., and Wine, P. H.: Chemical

20 Kinetics and Photochemical Data for Use in Atmospheric Studies, JPL Publication 10-06, Pasadena, California, 2011.

Saunders, S. M., Jenkin, M. E., Derwent, R. G., and Pilling, M. J.: Protocol for the development of the Master Chemical Mechanism, MCM v3 (Part A): tropospheric degradation of nonaromatic volatile organic compounds, Atmos. Chem. Phys., 3, 161–180, doi:10.5194/acp-3-161-2003, 2003.

Sauvage, B., Martin, R. V., van Donkelaar, A., Liu, X., Chance, K., Jaeglé, L., Palmer, P. I., Wu, S., and Fu, T.-M.: Remote sensed and in situ constraints on processes affecting tropical tropospheric ozone, Atmos. Chem. Phys., 7, 815–838, doi:10.5194/acp-7-815-2007, 2007. Singh, H. B., Salas, L., Herlth, D., Kolver, R., Czech, E., Avery, M., Crawford, J. H., Pierce, R. B.,

25

Sachse, G. W., Blake, D. R., Cohen, R. C., Bertram, T. H., Perring, A., Wooldridge, P. J., Dibb, J., Huey, G., Hudman, R. C., Turquety, S., Emmons, L. K., Flocke, F., Tang, Y., Carmichael, G. R., and Horowitz, L. W.: Reactive nitrogen distribution and partitioning in the



14308

North American troposphere and lowermost stratosphere, J. Geophys. Res.-Atmos., 112, D12S04, doi:10.1029/2006jd007664, 2007.

- Sprengnether, M., Demerjian, K. L., Donahue, N. M., and Anderson, J. G.: Product analysis of the OH oxidation of isoprene and 1,3-butadiene in the presence of NO, J. Geophys. Res.-Atmos., 107, 4269, doi:10.1029/2001jd000716, 2002.
- Atmos., 107, 4269, doi:10.1029/2001jd000716, 2002.
   Stavrakou, T., Muller, J. F., Boersma, K. F., De Smedt, I., and van der A, R. J.: Assessing the distribution and growth rates of NO<sub>x</sub> emission sources by inverting a 10-year record of NO<sub>2</sub> satellite columns, Geophys. Res. Lett., 35, L10801, doi:10.1029/2008gl033521, 2008.
  - Streets, D. G., Bond, T. C., Carmichael, G. R., Fernandes, S. D., Fu, Q., He, D., Klimont, Z.,
- Nelson, S. M., Tsai, N. Y., Wang, M. Q., Woo, J. H., and Yarber, K. F.: An inventory of gaseous and primary aerosol emissions in Asia in the year 2000, J. Geophys. Res.-Atmos., 108, 8809, doi:10.1029/2002jd003093, 2003.
  - Su, H., Cheng, Y. F., Oswald, R., Behrendt, T., Trebs, I., Meixner, F. X., Andreae, M. O., Cheng, P., Zhang, Y., and Poschl, U.: Soil Nitrite as a Source of Atmospheric HONO and OH Radicals, Science, 333, 1616–1618, doi:10.1126/science.1207687, 2011.
  - Thornton, J. A., Jaeglé, L., and McNeill, V. F.: Assessing known pathways for HO<sub>2</sub> loss in aqueous atmospheric aerosols: regional and global impacts on tropospheric oxidants, J. Geophys. Res.-Atmos., 113, D05303, doi:10.1029/2007jd009236, 2008.

15

25

Tost, H., Lawrence, M. G., Brühl, C., Jöckel, P., The GABRIEL Team, and The SCOUT-O3-

- <sup>20</sup> DARWIN/ACTIVE Team: Uncertainties in atmospheric chemistry modelling due to convection parameterisations and subsequent scavenging, Atmos. Chem. Phys., 10, 1931–1951, doi:10.5194/acp-10-1931-2010, 2010.
  - Valin, L. C., Russell, A. R., Hudman, R. C., and Cohen, R. C.: Effects of model resolution on the interpretation of satellite NO<sub>2</sub> observations, Atmos. Chem. Phys., 11, 11647–11655, doi:10.5194/acp-11-11647-2011, 2011.
  - Wang, S. W., Zhang, Q., Streets, D. G., He, K. B., Martin, R. V., Lamsal, L. N., Chen, D., Lei, Y., and Lu, Z.: Growth in NO<sub>x</sub> emissions from power plants in China: bottom-up estimates and satellite observations, Atmos. Chem. Phys., 12, 4429–4447, doi:10.5194/acp-12-4429-2012, 2012.
- Wu, S., Mickley, L. J., Jacob, D. J., Logan, J. A., Yantosca, R. M., and Rind, D.: Why are there large differences between models in global budgets of tropospheric ozone?, J. Geophys. Res.-Atmos., 112, D05302, doi:10.1029/2006jd007801, 2007.



Yang, F., Tan, J., Zhao, Q., Du, Z., He, K., Ma, Y., Duan, F., Chen, G., and Zhao, Q.: Characteristics of PM<sub>2.5</sub> speciation in representative megacities and across China, Atmos. Chem. Phys., 11, 5207–5219, doi:10.5194/acp-11-5207-2011, 2011.

Zhang, L., Jacob, D. J., Boersma, K. F., Jaffe, D. A., Olson, J. R., Bowman, K. W., Worden, J. R.,

- <sup>5</sup> Thompson, A. M., Avery, M. A., Cohen, R. C., Dibb, J. E., Flock, F. M., Fuelberg, H. E., Huey, L. G., McMillan, W. W., Singh, H. B., and Weinheimer, A. J.: Transpacific transport of ozone pollution and the effect of recent Asian emission increases on air quality in North America: an integrated analysis using satellite, aircraft, ozonesonde, and surface observations, Atmos. Chem. Phys., 8, 6117–6136, doi:10.5194/acp-8-6117-2008, 2008.
- <sup>10</sup> Zhang, Q., Streets, D. G., He, K., Wang, Y., Richter, A., Burrows, J. P., Uno, I., Jang, C. J., Chen, D., Yao, Z., and Lei, Y.: NO<sub>x</sub> emission trends for China, 1995–2004: the view from the ground and the view from space, J. Geophys. Res., 112, D22306, doi:10.1029/2007JD008684, 2007.

Zhang, Q., Streets, D. G., and He, K. B.: Satellite observations of recent power plant construc-

- tion in Inner Mongolia, China, Geophys. Res. Lett., 36, L15809, doi:10.1029/2009gl038984, 2009a.
  - Zhang, Q., Streets, D. G., Carmichael, G. R., He, K. B., Huo, H., Kannari, A., Klimont, Z., Park, I. S., Reddy, S., Fu, J. S., Chen, D., Duan, L., Lei, Y., Wang, L. T., and Yao, Z. L.: Asian emissions in 2006 for the NASA INTEX-B mission, Atmos. Chem. Phys., 9, 5131–5153, doi:10.5194/acp-9-5131-2009, 2009b.

20

Zhao, C. and Wang, Y. H.: Assimilated inversion of NO<sub>x</sub> emissions over east Asia using OMI NO<sub>2</sub> column measurements, Geophys. Res. Lett., 36, L06805, doi:10.1029/2008gl037123, 2009.

Diecuseion Pa	<b>ACPD</b> 12, 14269–14327, 2012							
ner I Die	Model uncertainties affecting NOx and ozone simulations							
clission	JT. Li	T. Lin et al.						
Dana	Title Page							
	Abstract	Introduction						
_	Conclusions	References						
	Tables	Figures						
		▶1						
ים חם	•	•						
_	Back	Close						
	Full Screen / Esc							
	Printer-frien	dly Version						
עס	Interactive Discussion							
D								

#### Table 1. Sensitivity simulations by adjusting model meteorological or chemical parameters <sup>1</sup>

Case	Parameter	Adjustments	Notes & ref.
1	Base simulation	No changes in parameters	As in Lin (2012)
2	Air temperature in the lowest 10 layers (about 0–1.3 km)	Decreased by 2 °C in the daytime and increased by 1 °C at night	Based on ISH data
3 <sup>2</sup>	Air temperature in the lowest 10 layers (about 0–1.3 km)	Decreased by 5 °C in the daytime and increased by 1 °C at night	Based on ISH data
4	Air temperature in the lowest four layers (about 0–0.5 km)	Decreased by 5 °C in the daytime and increased by 1 °C at night	Based on ISH data
5	Air temperature in the lowest 10 layers (about 0–1.3 km)	Increased by 5 °C at night	Based on ISH data
6	Tropospheric water vapor content	Scaled up by 130 %	Based on ISCCP data
7 <sup>2</sup>	Tropospheric water vapor content	Scaled down by 130 %	Based on ISCCP data
8	Cloud optical depth	Assumed to be zero	Based on ISCCP data
9	Cloud optical depth	Decreased by 50 %	Based on ISCCP data
10	Cloud optical depth	Increased by 50 %	Based on ISCCP data
11 <sup>2</sup>	Cloud optical depth	Increased by 100 %	Based on ISCCP data
12	Cloud optical depth	Evenly distributed in all tropospheric layers	Based on ISCCP data and Liu et al. (2009)
13	PBLH	Derived online	
14	PBLH	Decreased by 10 % in the daytime and tripled at night	Based on ISH temperature data and Liu et al. (2010)
15	PBLH	Inpled at night	and Liu et al. (2010)
16	PBLH	PBLH taken from GEOS-4	
17	PBLH	Scaled up by 130 %	To analyze potential causes of underestimated NO <sub>2</sub> columns
18	PBLH	Scaled up by 200 %	To analyze potential causes of underestimated NO <sub>2</sub> columns
19 <sup>-2</sup>	Rate constant for OH + NO <sub>2</sub> reaction	Scaled down by 130 %	Mollner et al. (2010); Sander et al. (2011)
20	Rate constant for OH + NO <sub>2</sub> reaction	Scaled up by 130 %	Mollner et al. (2010); Sander et al. (2011)
21	Yield of HNO <sub>3</sub> from NO + HO <sub>2</sub> reaction	Assumed to be 5%	Butkovskaya et al. (2005, 2007, 2009); Sander et al. (2011)
22	Yield of HNO <sub>3</sub> from NO + HO <sub>2</sub> reaction	Assumed to be 1 %	Butkovskaya et al. (2005, 2007, 2009); Sander et al. (2011)
23 <sup>2</sup>	OH regeneration from isoprene chemistry	Yield of 100% from isoprene + OH reaction as- sumed	Butler et al. (2008); Lelieveld et al. (2008); Kubistin et al. (2010)
24	OH regeneration from isoprene chemistry	Isoprene + OH reaction turned off	Kubistin et al. (2010)
25 <sup>2</sup>	Net yield of isoprene nitrates	Decreased from 10 % to 2.4 %	See review by Paulot et al. (2009). Transport of isoprene nitrates is neglected.
26	Equilibrium constant of PAN	Decreased by 20 %	Sander et al. (2011)
27 <sup>2</sup>	Uptake rate of HO <sub>2</sub> on aerosols	Set at 0.2	Thornton et al. (2008); Jacob (2000); He et al. (2001); Yang et al. (2011)
28	Uptake rate of HO <sub>2</sub> on sulfate-nitrate-ammonium aerosols in the continental PBL	Set at 0.2	Thornton et al. (2008); Jacob (2000); He et al. (2001); Yang et al. (2011)
29 <sup>2</sup>	Uptake rate of N <sub>2</sub> O <sub>5</sub> on aerosols	Scaled down by a factor of 10	Bertram et al. (2009); Brown et al. (2009)
30 <sup>2</sup>	Aerosol scattering and absorption	Turned off for all aerosols	Based on comparison with MODIS AOD
31	Aerosol scattering and absorption	Turned off for BC alone	Based on comparison with MODIS AOD
32	Emissions of CO and SO <sub>2</sub>	Increased by 50 %	Following Lin et al. (2010b)
33	Emissions of propene	Increased by 50 %	To represent effects of aromatics based on INTEX-
	a property and	, · ·	B inventory (Zhang et al., 2009)
34 <sup>2</sup>	Emissions of propene	Increased by 300 %	To represent effects of aromatics based on top- down constraint (Liu et al., 2010, 2011)
35	Emissions of CO, $\mathrm{SO}_{\mathrm{2}}$ and VOC	Increased by 50 %	Following Lin et al. (2010b)

<sup>1</sup> Analyses are done in Sects. 4 and 5 also for other parameters without conducting additional sensitivity simulations.

<sup>2</sup> These cases are used for post-model modifications described in Table 2 and analyzed in Sects. 6 and 7.



14310

# **Table 2.** Parameters considered in post-model modification for $NO_2$ columns and surface ozone at the 284 meteorological stations in East China.

Parameter	Modifications	Case in Table 1 <sup>1</sup>	Impacts on NO <sub>2</sub> columns <sup>2</sup>	Impacts on surface ozone <sup>2</sup>	Notes & ref.
(a) Air temperature	Based on ISH values	3	Insignificant	Large in July and small in January	Modification is assumed for the lowest 10 layers
(b) Tropospheric water vapor content	Based on ISCCP values	7	Small in July and insignificant in January	Large	
(c) Cloud optical depth	Based on ISCCP values	11	Very large in July and insignifi- cant in January	Large in July and very large in January	
(d) Rate constant of OH + NO <sub>2</sub> reaction	lowered by 15 %	19	Large in July and small in Jan- uary	Small	Mollner et al. (2010); Sander et al. (2011)
(e) OH regeneration from iso- prene chemistry	Yield of 100 % from isoprene + OH reaction assumed	23	Small in July and insignificant in January	Insignificant	Butler et al. (2008); Lelieveld et al. (2008); Kubistin et al. (2010)
(f) Net yield of isoprene nitrates	6 % instead of 10 %	25	insignificant	Insignificant	See review by Paulot et al. (2009)
(g) Uptake rate of HO <sub>2</sub> on aerosols	0.2 instead of the default setup; biases in model AOD with re- spect to MODIS accounted for	27	Very large	Very large	Thornton et al. (2008); Jacob, (2000); He et al. (2001); Yang et al. (2011)
(h) Uptake rate of $N_2 O_5$ on aerosols	10% of the default setup; bi- ases in model AOD with respect to MODIS accounted for	29	Insignificant in July and large in January	Small in July and in- significant in January	Bertram et al. (2009); Brown et al. (2009)
(i) Aerosol scattering and ab- sorption	Scaled based on MODIS AOD in comparison with model AOD	30	Insignificant	Insignificant in July and large in January	
(j) Emissions of propene	Increased by 300 %	34	Insignificant in July and large in January	Large in July and very large in January	To represent effects of aromatics based on top-down constraint (Liu et al., 2010, 2011)

<sup>1</sup> Results from sensitivity simulations described in Table 1 are used as the basis of respective post-model modifications here, i.e., they provide  $\partial \ln \Omega / \partial P_i$  in Eq. (1) for individual parameters.

<sup>2</sup> Impacts with the slopes in Figs. 13–16 deviating from unity by more than 10%, 5–10%, 2–5%, and within 2% are indicated here as very large, large, small, and insignificant, respectively. For ozone, impacts are indicated also as large (very large) if the  $R^2$  in Figs. 15, 16 is reduced to about 0.90–0.95 (lower than 0.90). Note that these indications are for the overall impact in East China; and impacts may be larger or smaller at individual stations.



Discussion Paper

**Discussion** Paper

**Discussion** Paper

**Discussion** Paper





**Fig. 1.** Tropospheric chemistry involving NO<sub>x</sub> and impacts of meteorological and chemical parameters evaluated in the present study. Processes shown in solid grey arrows are discussed without sensitivity simulations. Processes shown in dashed grey arrows are not discussed explicitly. Note that PBL mixing and convection affect vertical distributions of NO<sub>x</sub> and related species. Heterogeneous uptake on aerosols depends on the amount of aerosol surfaces as well. Evaluation on the RONO<sub>2</sub> pathway is focused mainly on isoprene nitrates. Clouds and water vapor have indirect influences on radicals through effects on solar radiation.











**Fig. 3.** Spatial distribution of percentage differences between modeled NO<sub>2</sub> columns with and without adjustments in air temperature: the daytime temperature is decreased by  $5^{\circ}$ C with an increase of 1 °C at night for the lowest 10 model layers. See Fig. S9 for more results of sensitivity tests on temperature.











**Fig. 5.** Spatial distribution of percentage differences between modeled  $NO_2$  columns with and without doubling the COD. See Fig. S10 for more results of sensitivity tests on COD.











**Fig. 7.** Spatial distribution of percentage differences between modeled NO<sub>2</sub> columns with and without scaling down the rate constant of  $OH + NO_2$  reaction by a factor of 130 %.





**Fig. 8.** Spatial distribution of percentage differences between modeled NO<sub>2</sub> columns with and without assuming a yield of HNO<sub>3</sub> at 5 % (**a**, **b**) or 1 % (**c**, **d**) from the HO<sub>2</sub> + NO reaction.























**Fig. 12.** Spatial distribution of percentage differences between modeled  $NO_2$  columns with and without increasing emissions of propene by 300 %. See Fig. S16 for more results of sensitivity tests on non-NO<sub>x</sub> emissions.











Fig. 14. Similar to Fig. 13 but for January 2006.





Fig. 15. Similar to Fig. 13 but for surface ozone (ppb) in July 2006.





Fig. 16. Similar to Fig. 13 but for surface ozone (ppb) in January 2006.

